

Optical Atomic Clocks



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The only reason for time is so that everything doesn't happen at once.

- *Albert Einstein*

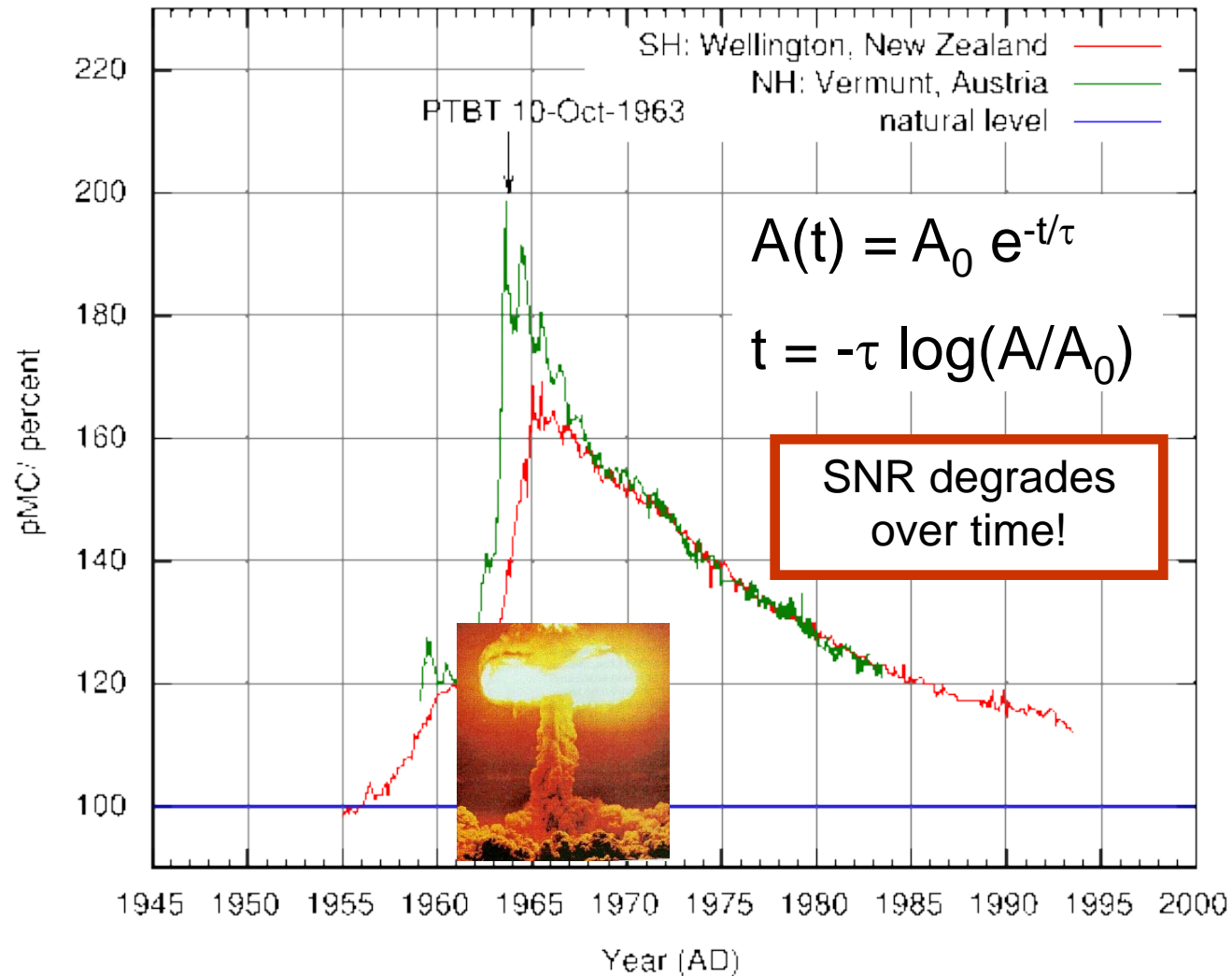
What is a clock?



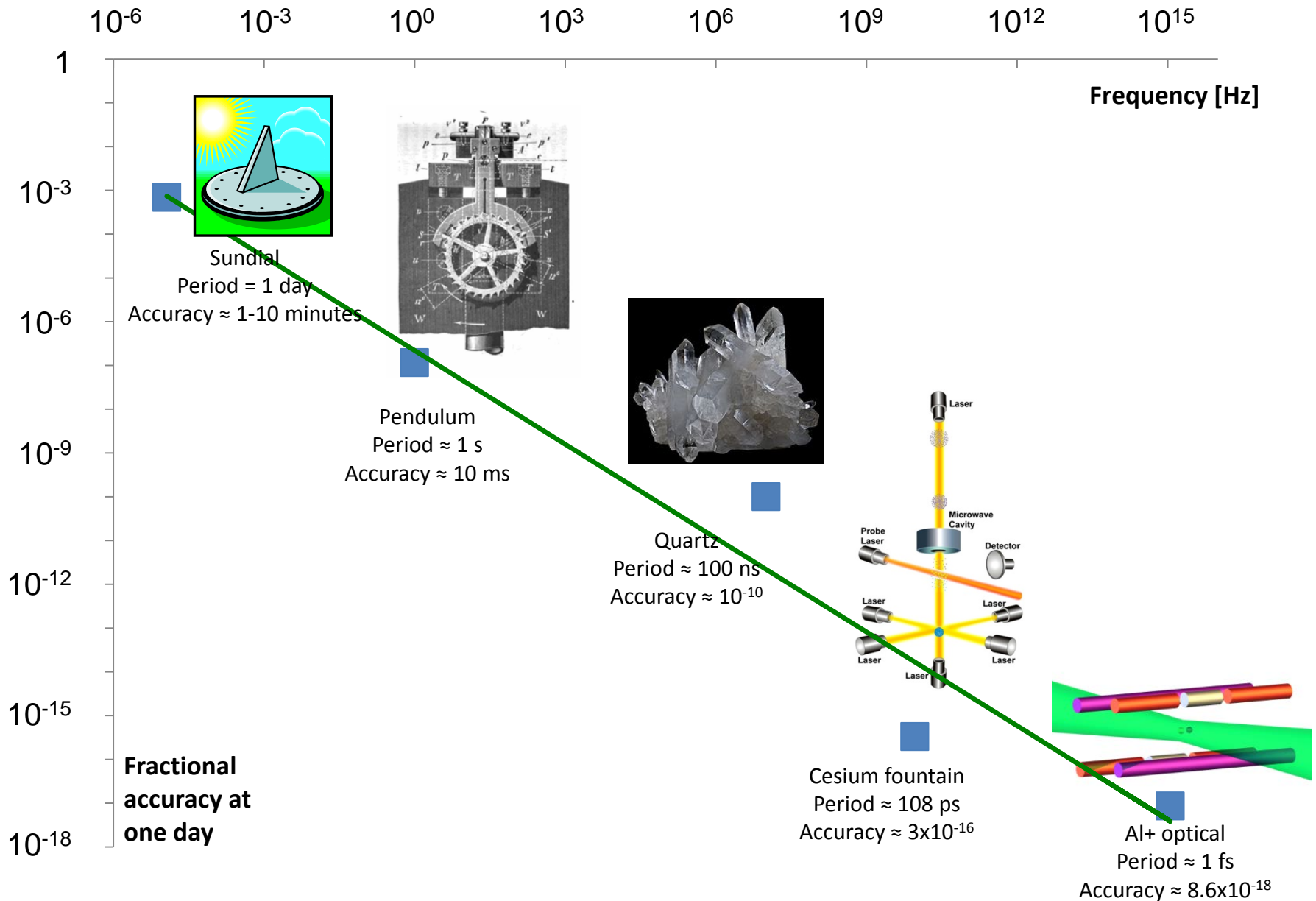
Salvador Dalí: *The Persistence of Memory* (1931)

*Any time-dependent
physical process that can be
inverted mathematically*

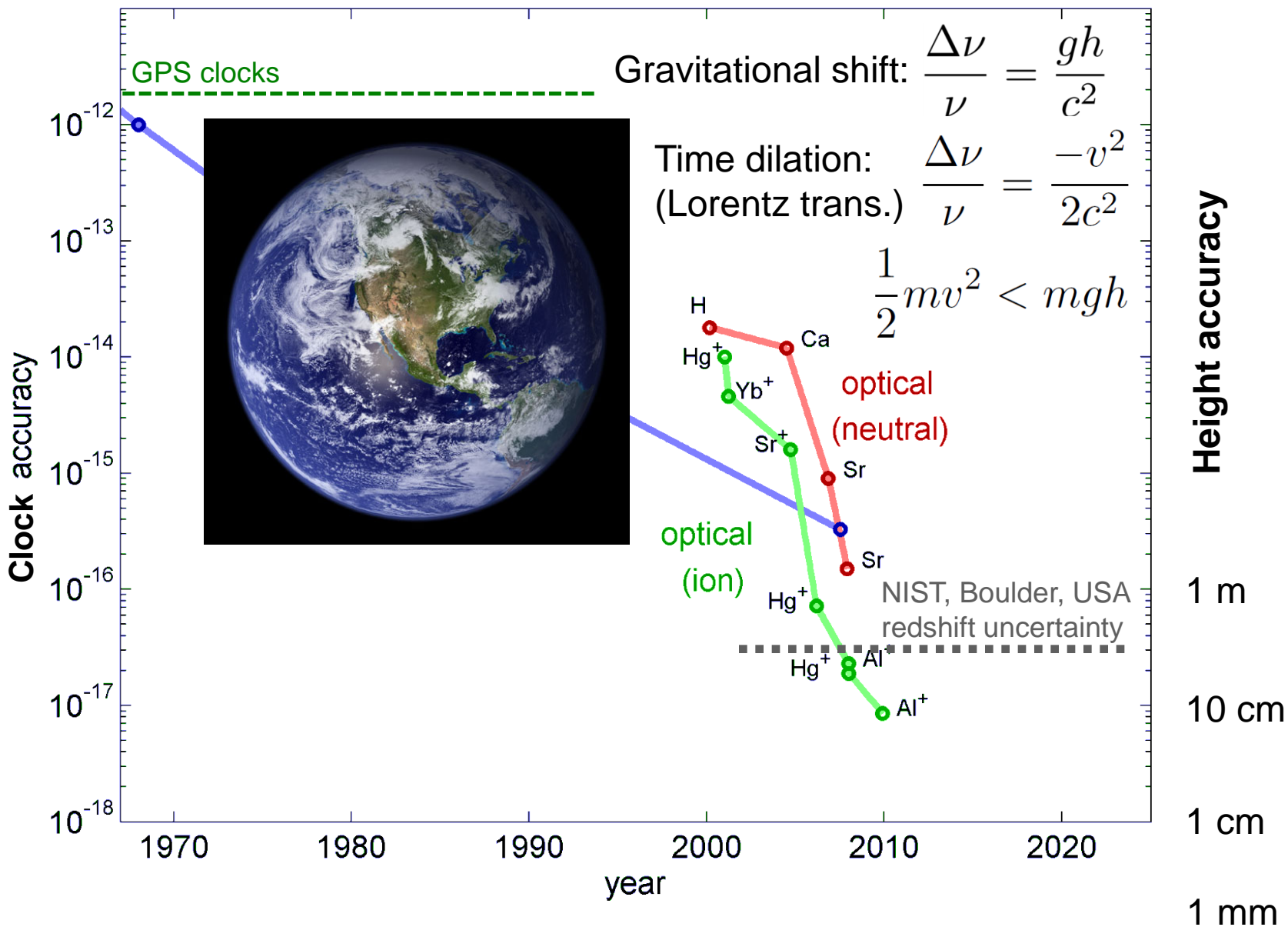
Radiocarbon clock



Clock speed vs. accuracy



Recent atomic clock accuracy



$^{27}\text{Al}^+$

$I = 5/2$

$2p^6 3s 3p$

3P_0 $\lambda=267.43$ nm, $\tau=21$ s

$2p^6 3s^2$ 1S_0



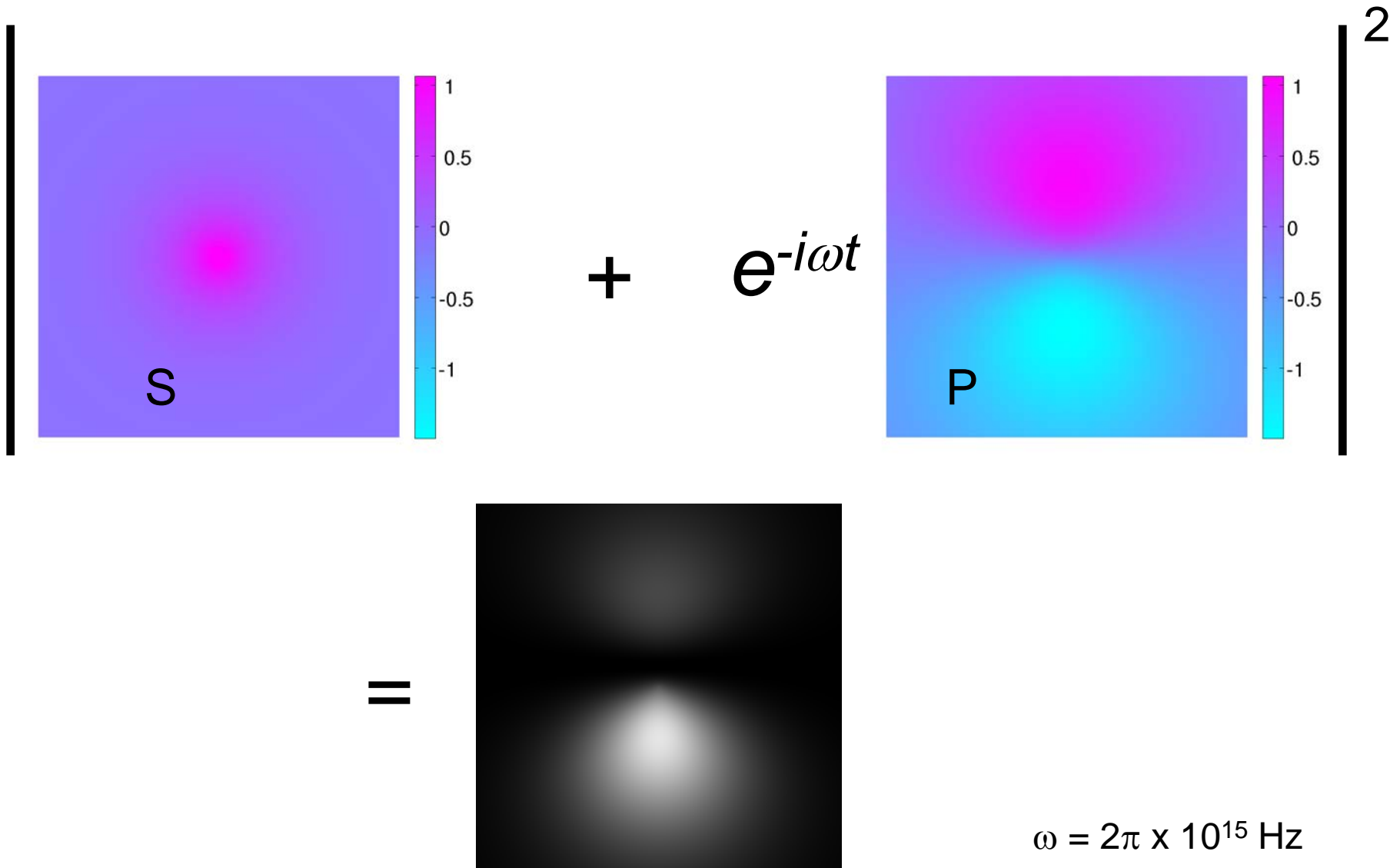
$$E = 4.636\,146\,95\,(10)\text{ eV}$$

$$f = 1\,121\,015\,393\,207\,857.4(7)\text{ Hz}$$

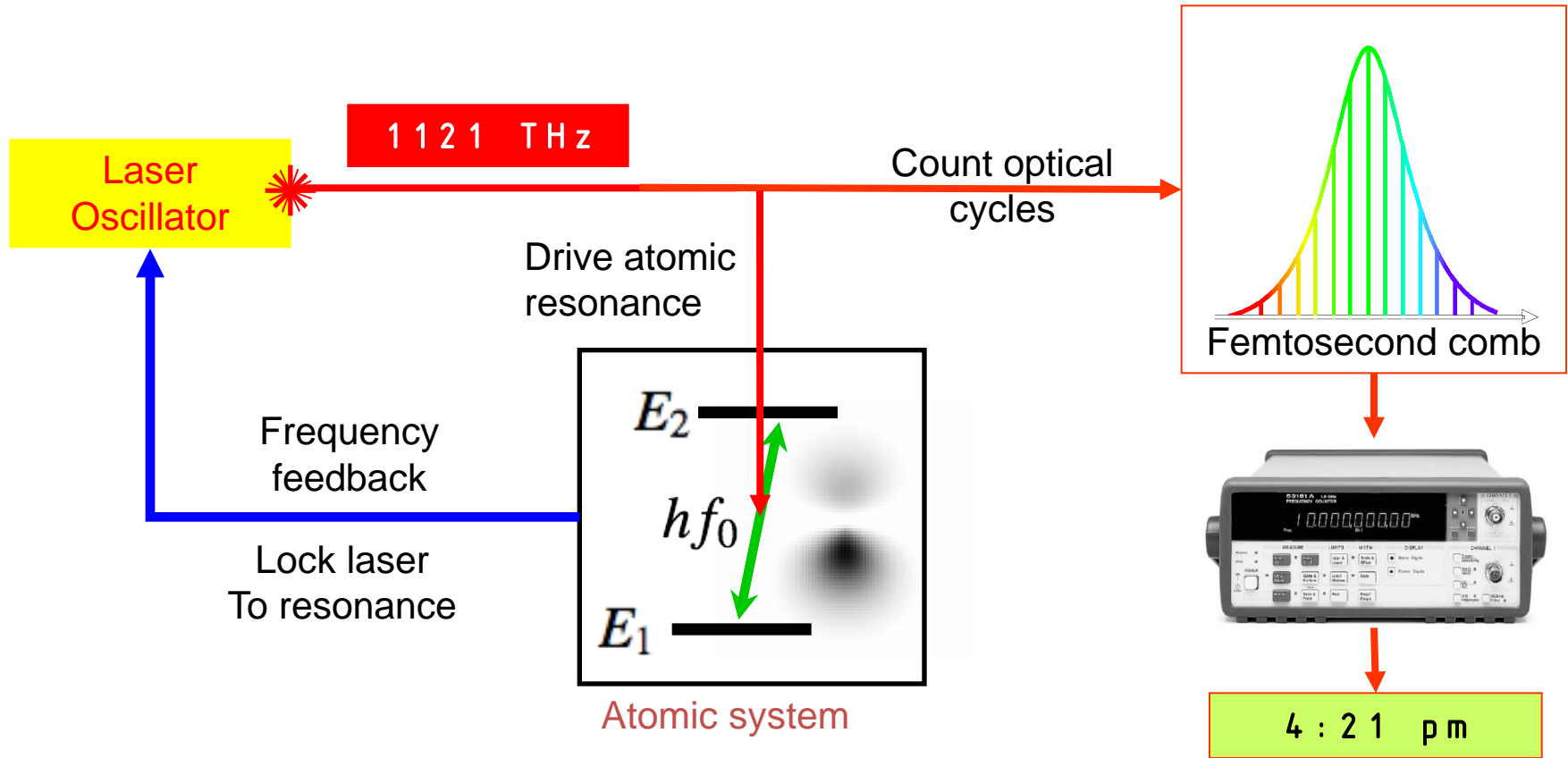
- Insensitive to external fields
- Smallest known temperature sensitivity (5×10^{-20} / Kelvin)
- Two-ion quantum logic techniques for cooling, state preparation & clock readout [1]

[1] D.J. Wineland *et al.*,
Proc. 6th Symp. Freq. Stds. and
Metrology, 2001, pp. 361-368

“Ticking” of Al⁺ clock

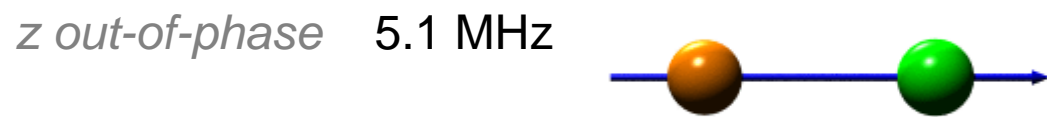
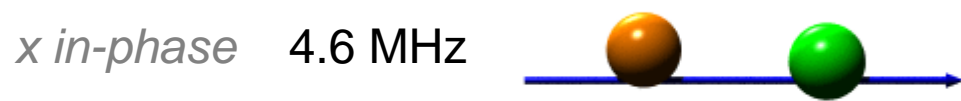
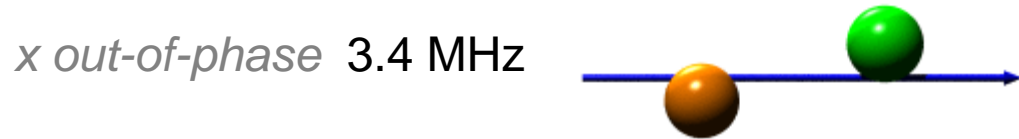
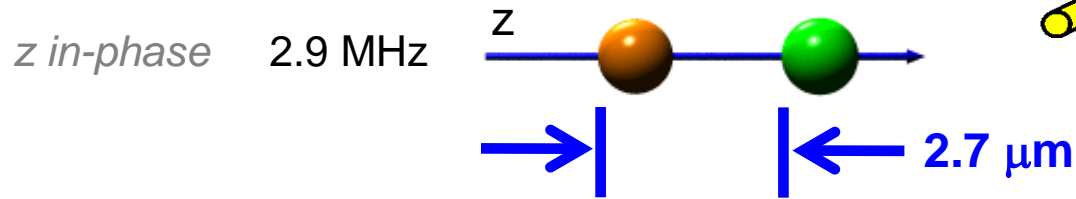
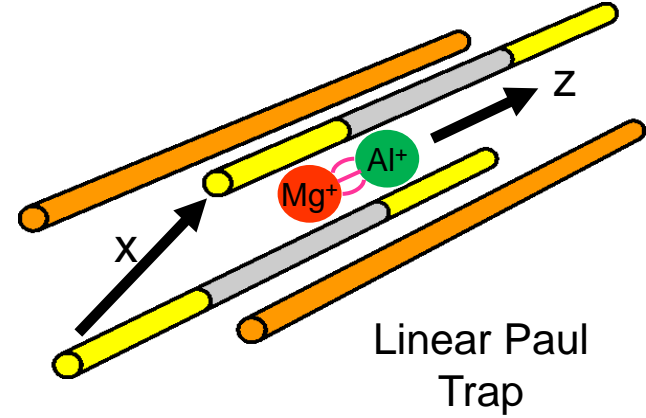


Optical Atomic Clock



Clock frequency: $f_0 = \frac{E_2 - E_1}{h} \approx 10^{15} \text{ Hz}$

Coupled motion (normal modes)



Time dilation:

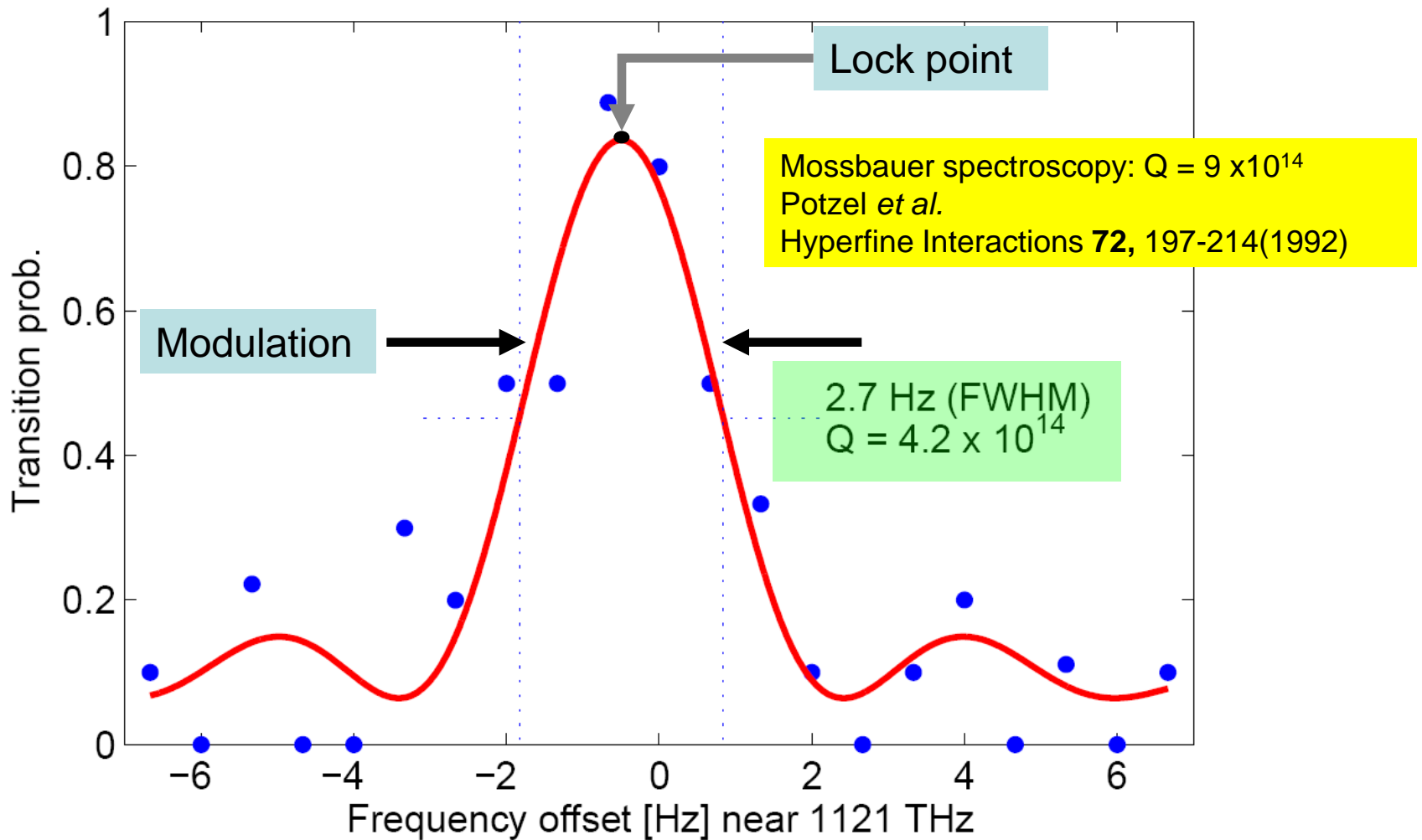
$$\delta v/v = v^2/(2c^2) \quad mv^2 = kT$$

$$T = 300\text{K} \rightarrow \delta v/v = 5 \times 10^{-11}$$

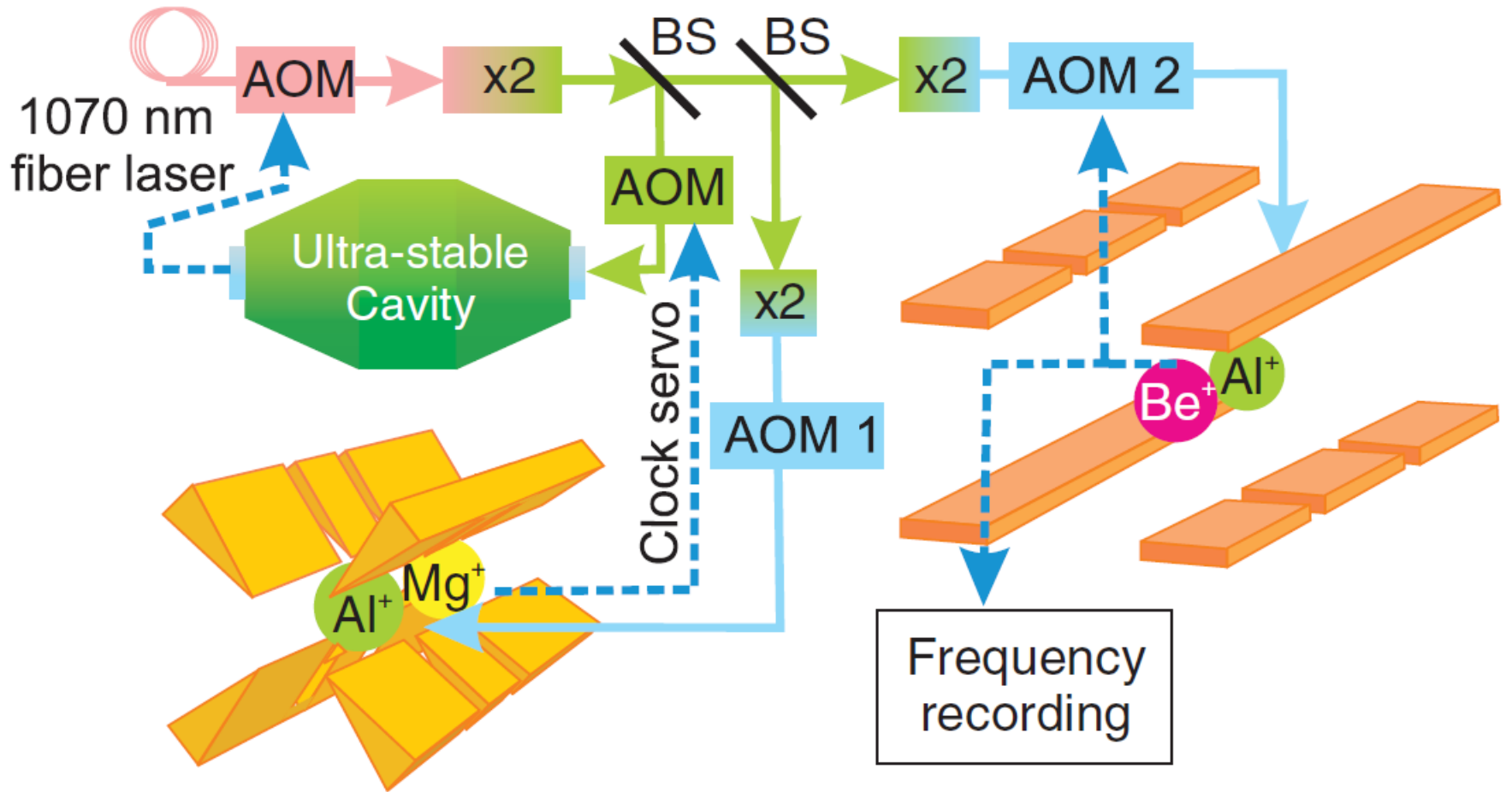
$$T = 0.0001\text{K} \rightarrow \delta v/v = 10^{-18}$$

Al⁺ Spectroscopy

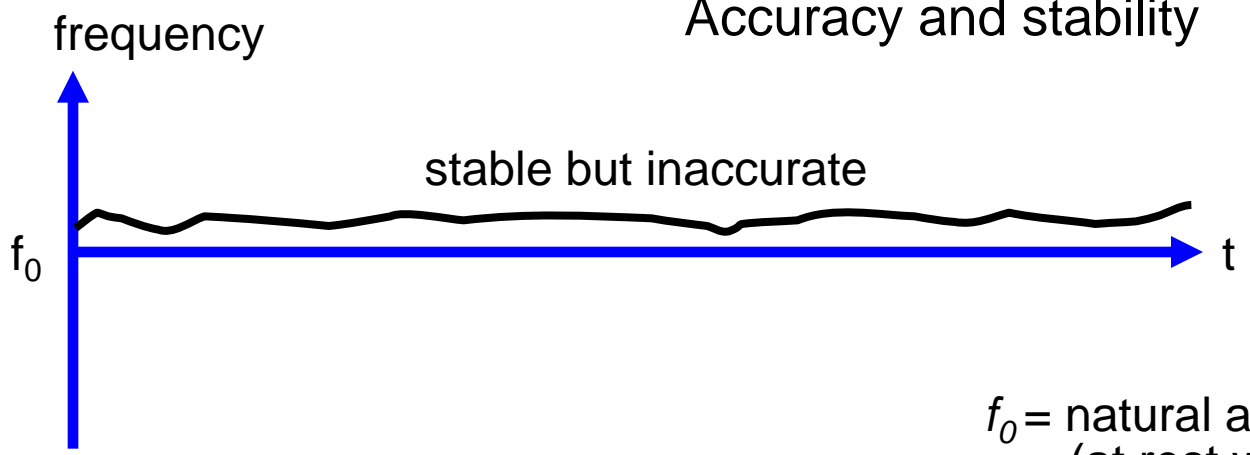
Al⁺ $^1S_0 - ^3P_0$ resonance (10 scans, 300 ms probe time)



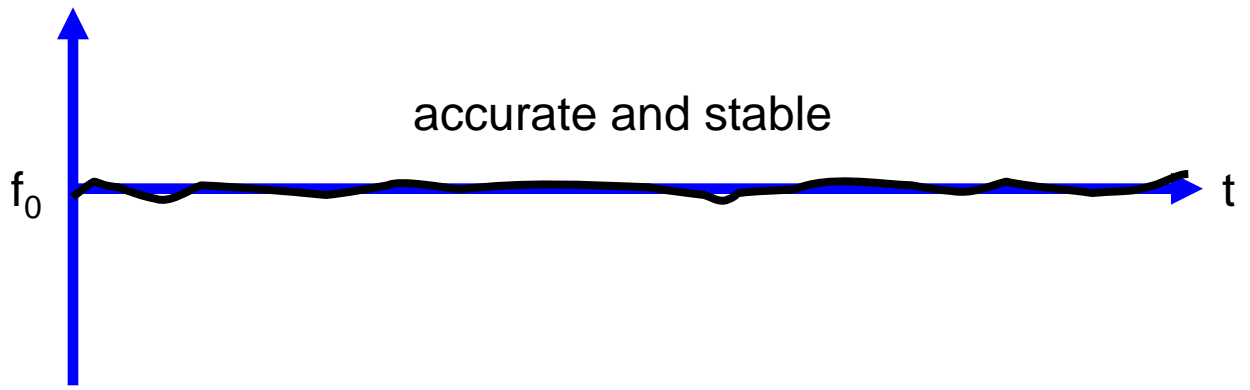
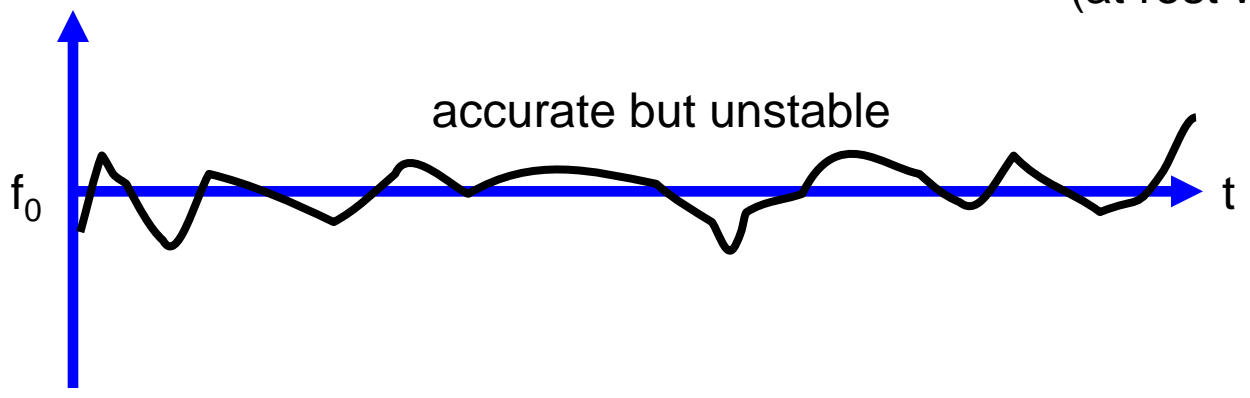
$^{27}\text{Al}^+$ vs. $^{27}\text{Al}^+$



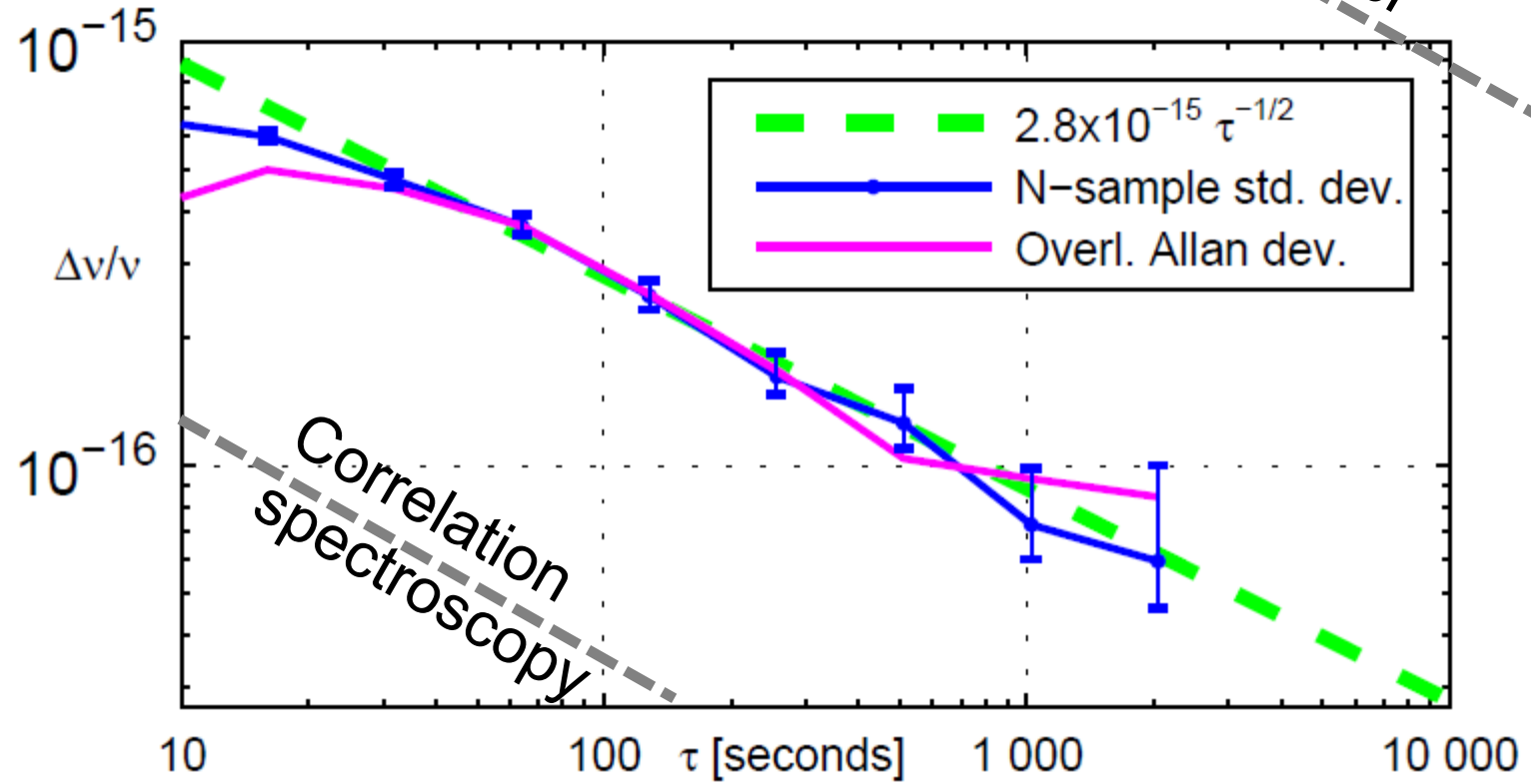
Accuracy and stability



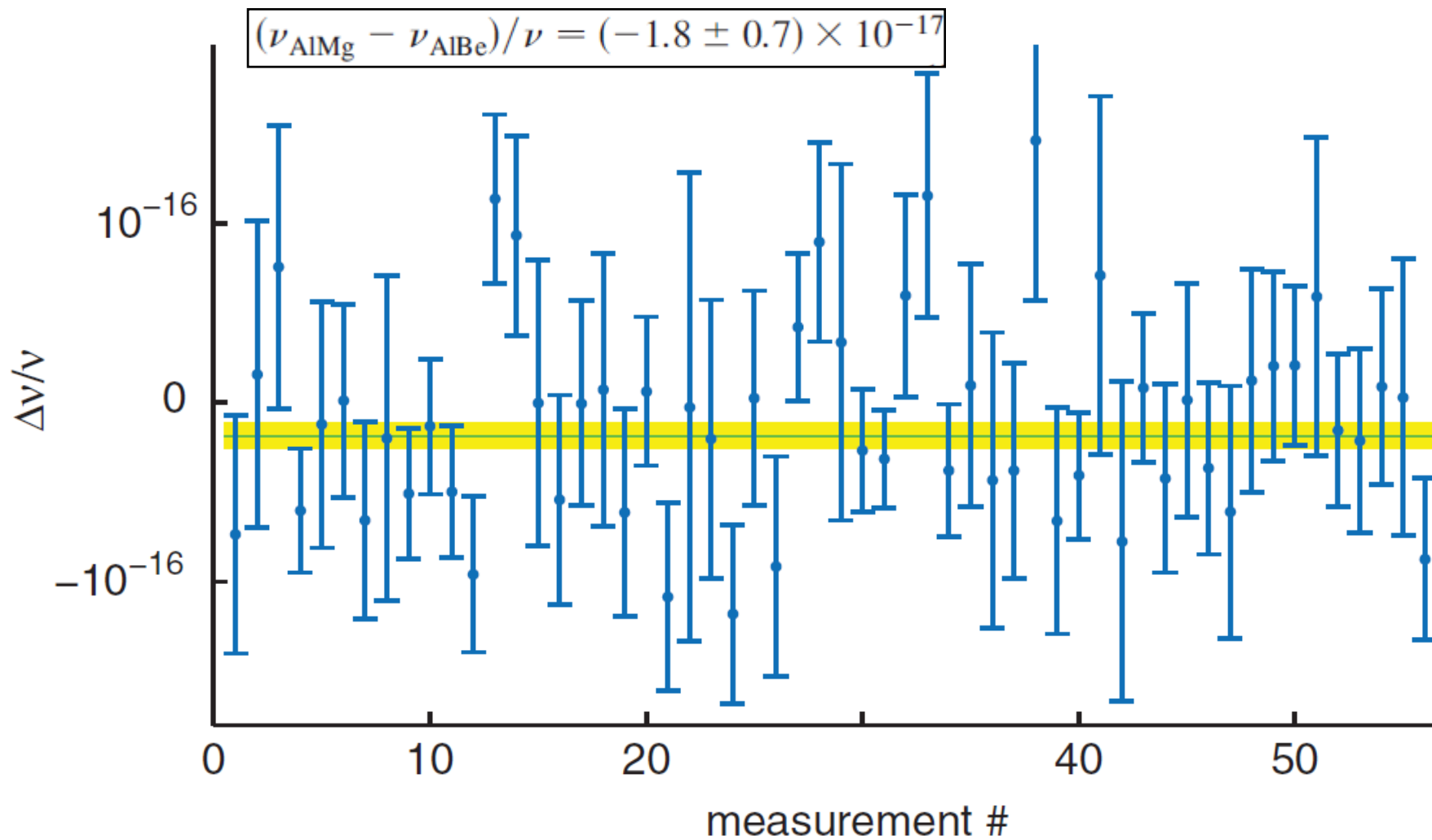
$f_0 =$ natural atomic resonance frequency
(at rest without perturbations)



$^{27}\text{Al}^+$ vs. $^{27}\text{Al}^+$ stability



$^{27}\text{Al}^+$ vs. $^{27}\text{Al}^+$

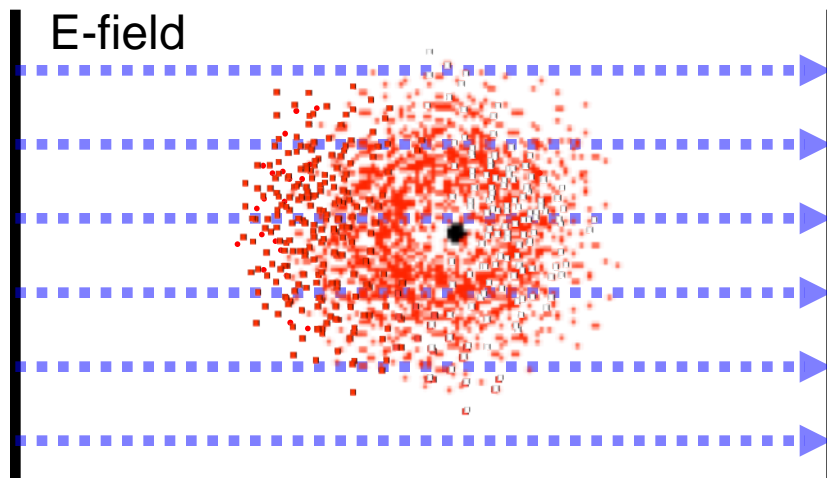
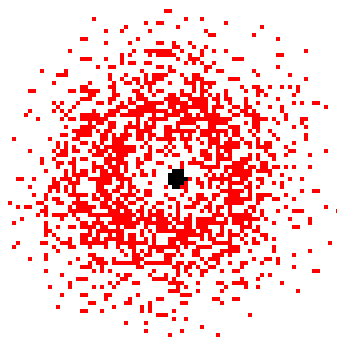


$^{27}\text{Al}^+$ error budgets

Reduced in 3rd gen. trap ?

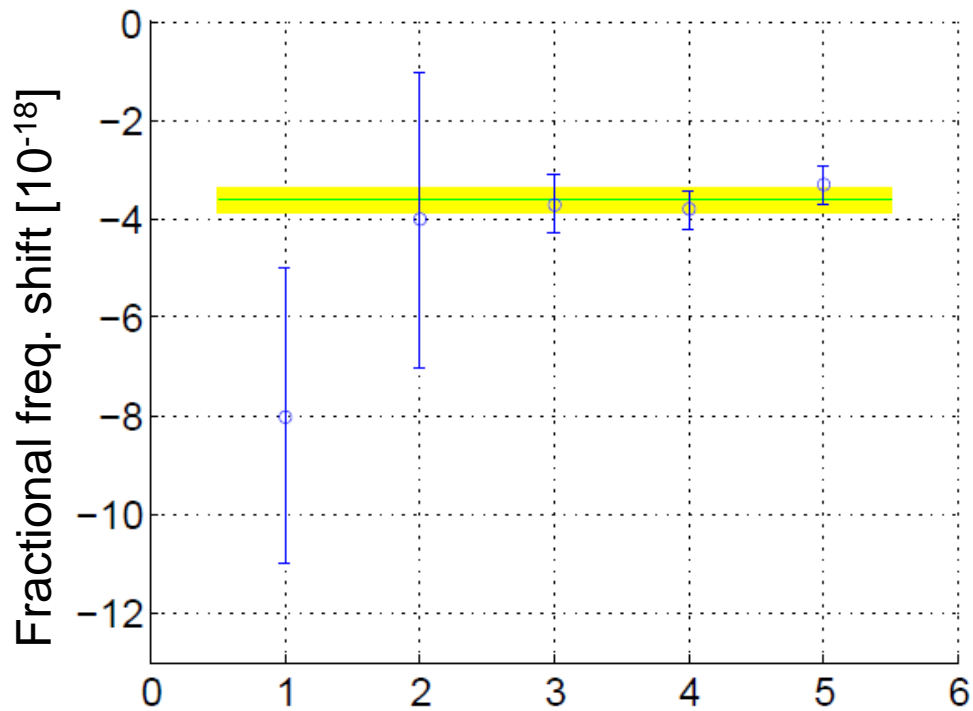
Effect	Parameter	Al ⁺ /Be ⁺	Al ⁺ /Mg ⁺
		1 st gen. uncertainty [x 10 ⁻¹⁷]	2 nd gen. uncertainty [x 10 ⁻¹⁷]
Blackbody radiation shift	Operating temperature/ Polarizability	0.5	0.3
Micromotion time dilation	Axial RF fields (trap imperfections)	0.3	0.6
Micromotion time dilation	Radial static field (stray charges)	2	
Secular motion time dilation	Radial temperature (measurement uncertainty)	0.8	0.5
2 nd order Zeeman	RMS magnetic field	0.1	0.07
Cooling laser Stark shift	I / Isat and Polarizability	0.2	0.15
1 st -order Doppler from correlated ion movement		0.1	.03
Total (quadrature sum)		2.3	0.86

Room Temperature Blackbody shift



Quadratic Stark shift

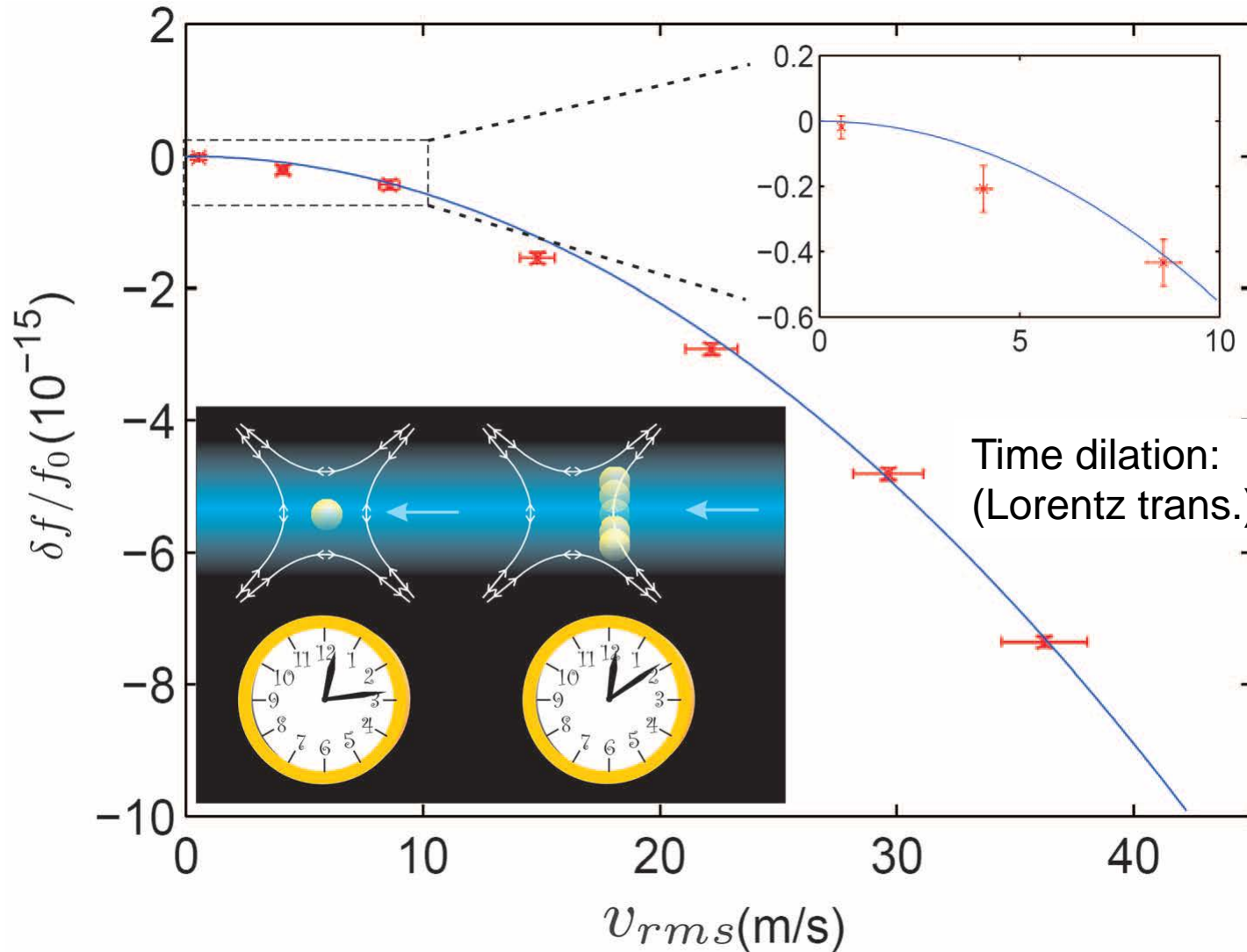
Room Temperature Blackbody shift



- [1] Rosenband et al. 2006 $-8 \pm 3 \times 10^{-18}$
- [2] Mitroy et al. 2008 $-4 \pm 3 \times 10^{-18}$
- [3] Kallay et al. 2011 $-3.7 \pm 0.6 \times 10^{-18}$
- [4] Safronova et al. 2011 $-3.8 \pm 0.4 \times 10^{-18}$
- [5] In preparation $-3.3 \pm 0.4 \times 10^{-18}$

Species	fractional shift [$\times 10^{-18}$]
Al ⁺	3.7 +/- 0.4
In ⁺	< 70
Yb ⁺	580
Sr ⁺	670
Ca	2200
Yb	2400
Sr	5100
Cs	21000

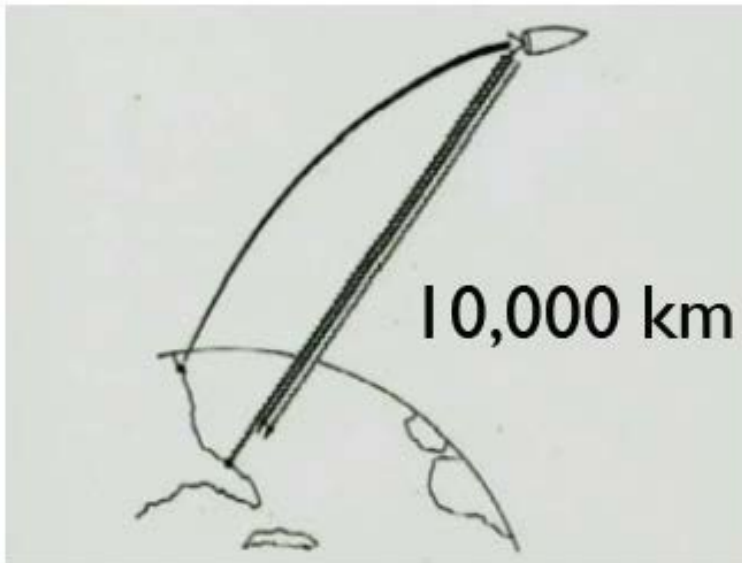
Twin paradox (motional time-dilation)



$$\frac{\Delta\nu}{\nu} = \frac{-v^2}{2c^2}$$

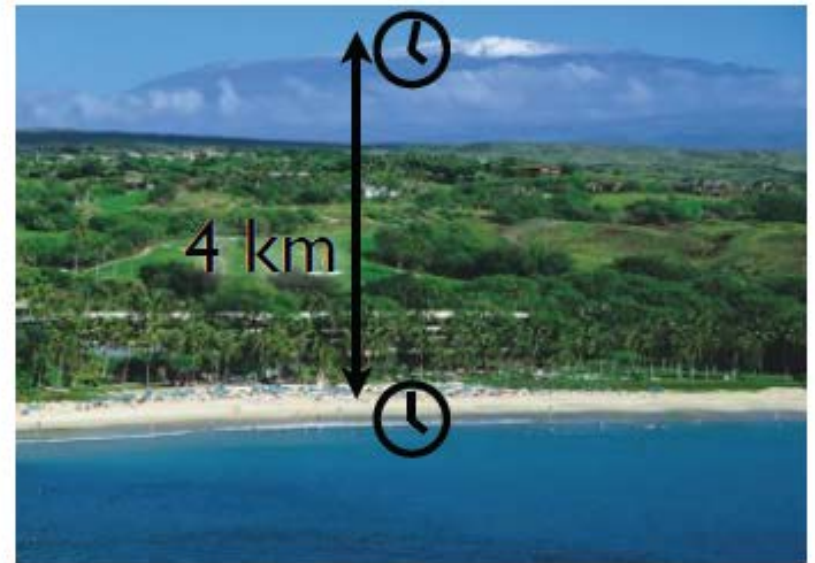
Gravitational redshift measurements

$$\frac{\Delta\nu}{\nu} = \frac{gh}{c^2}$$



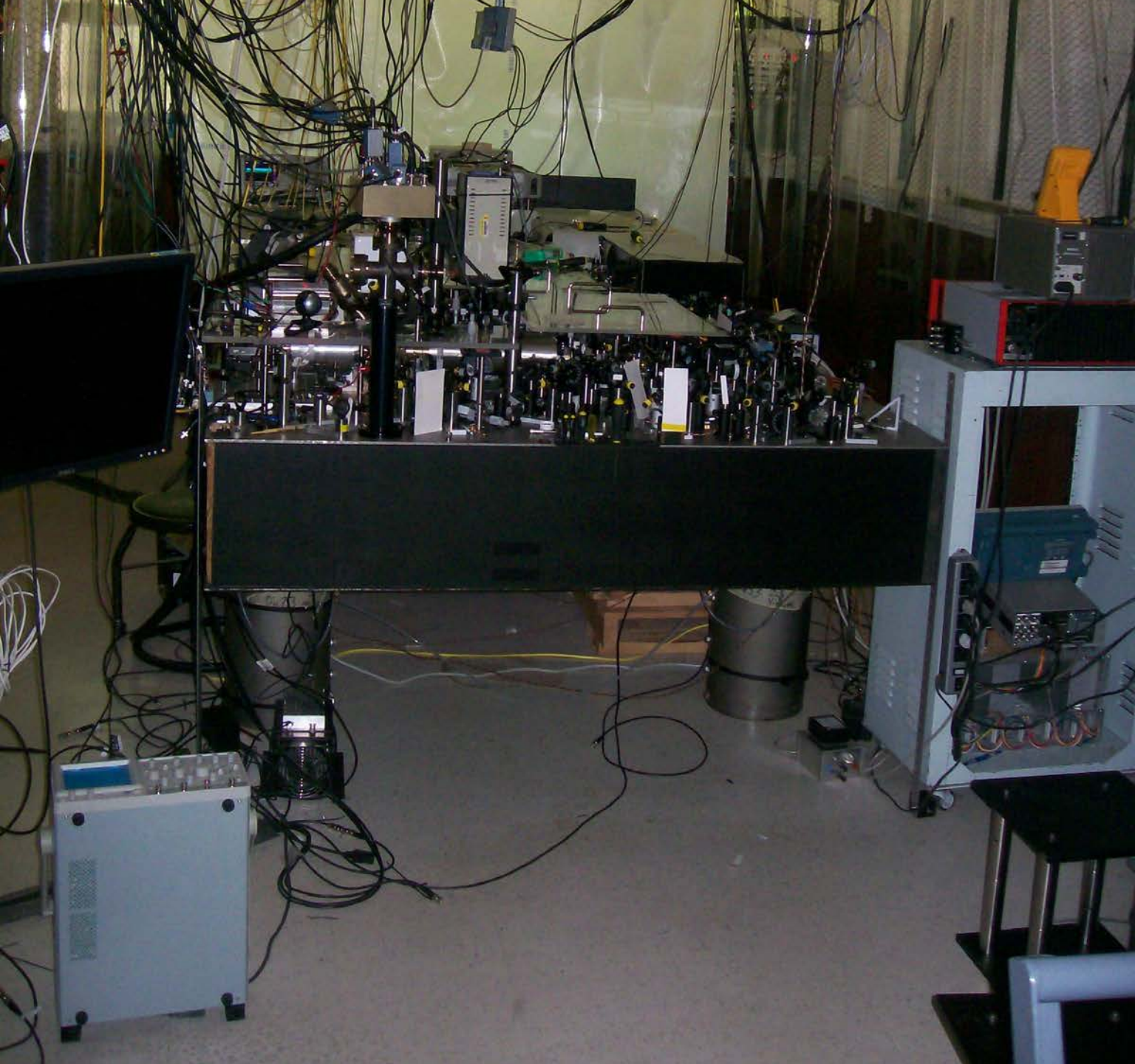
1976

Vessot et al., maser on rocket
Test of redshift with uncertainty 10^{-4}



2017?

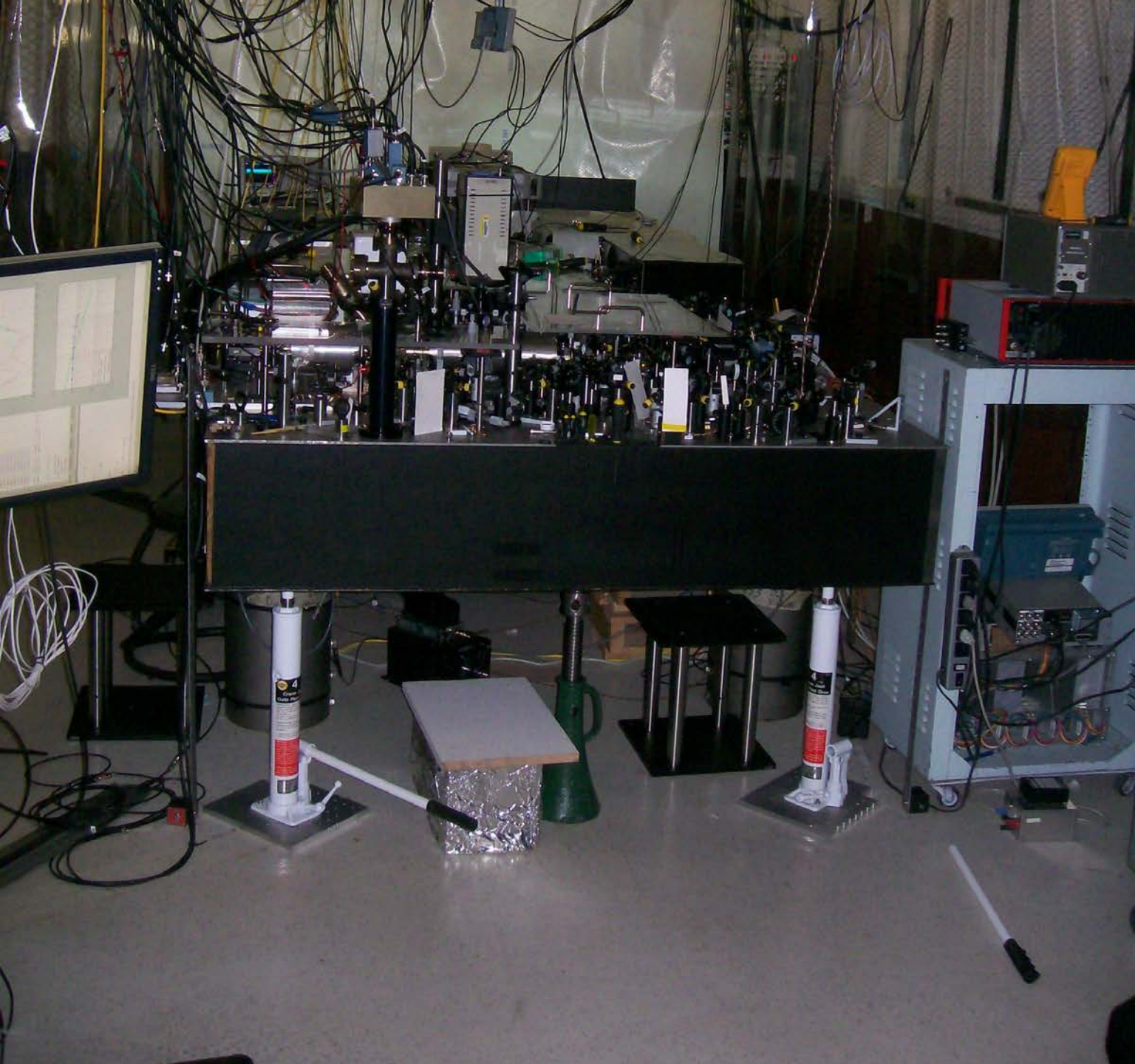
Comparison of Al+ clocks
Redshift uncertainty $< 3 \times 10^{-5}$



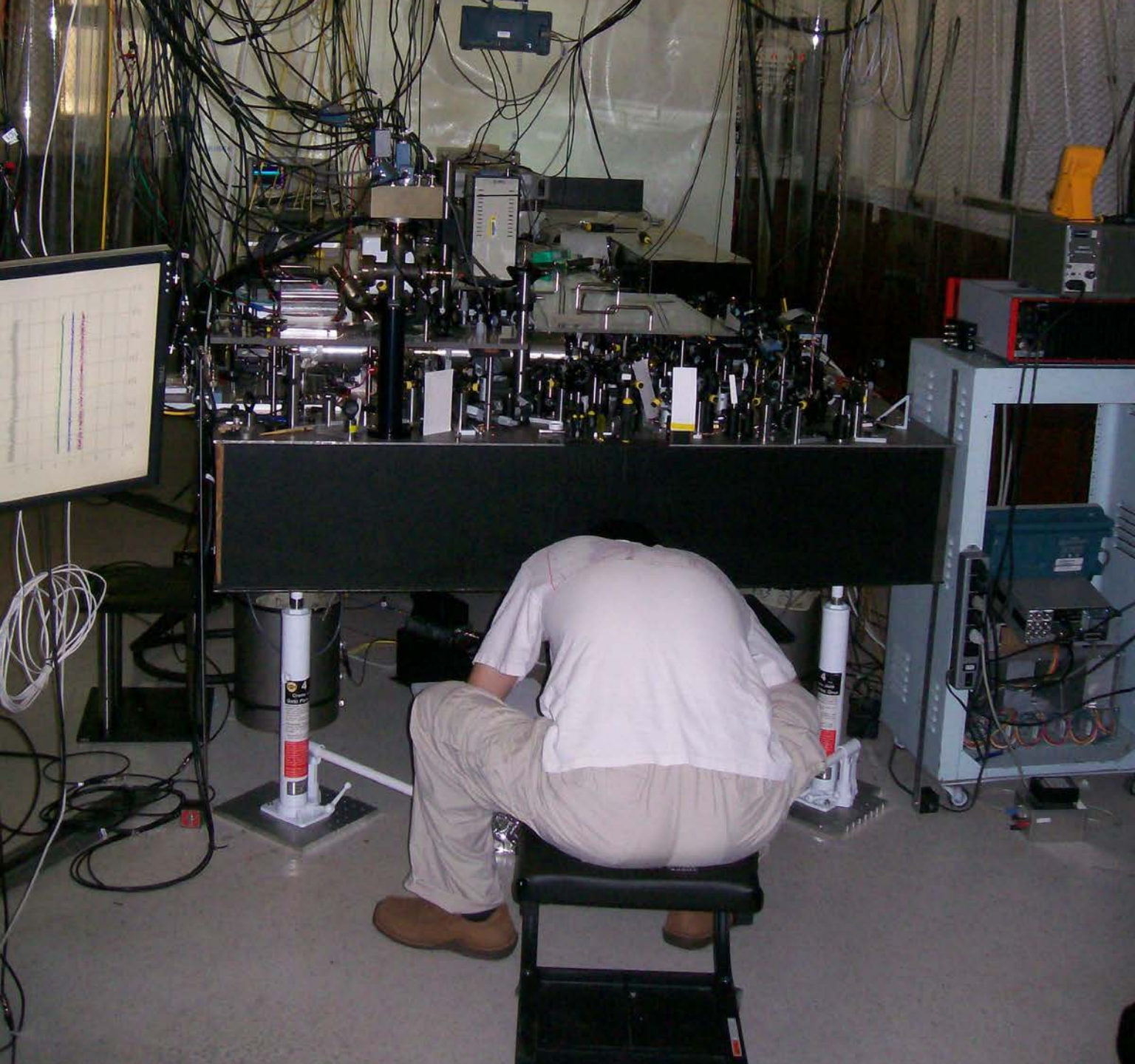
0×10^{-18}



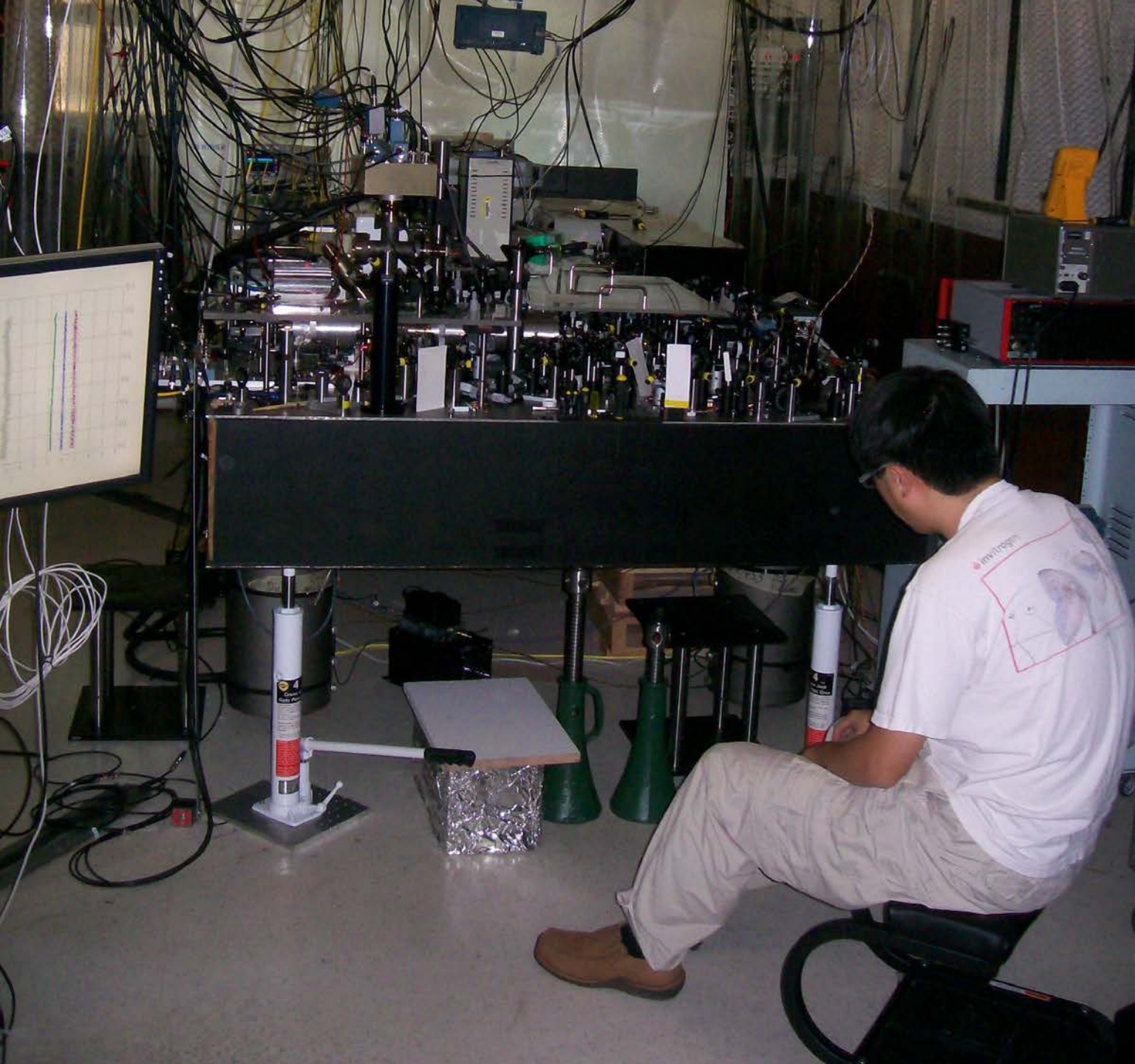
0×10^{-18}



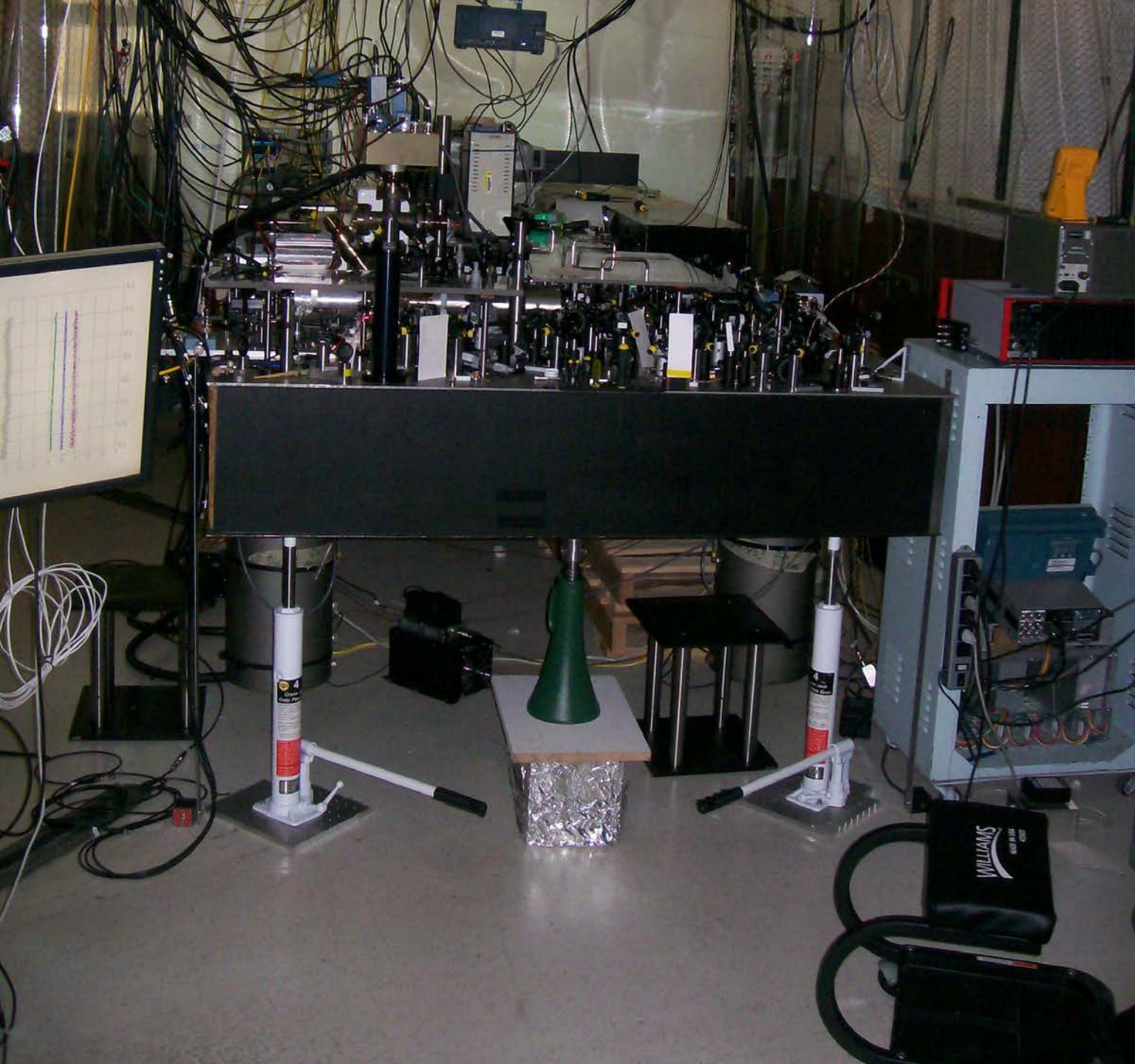
3×10^{-18}



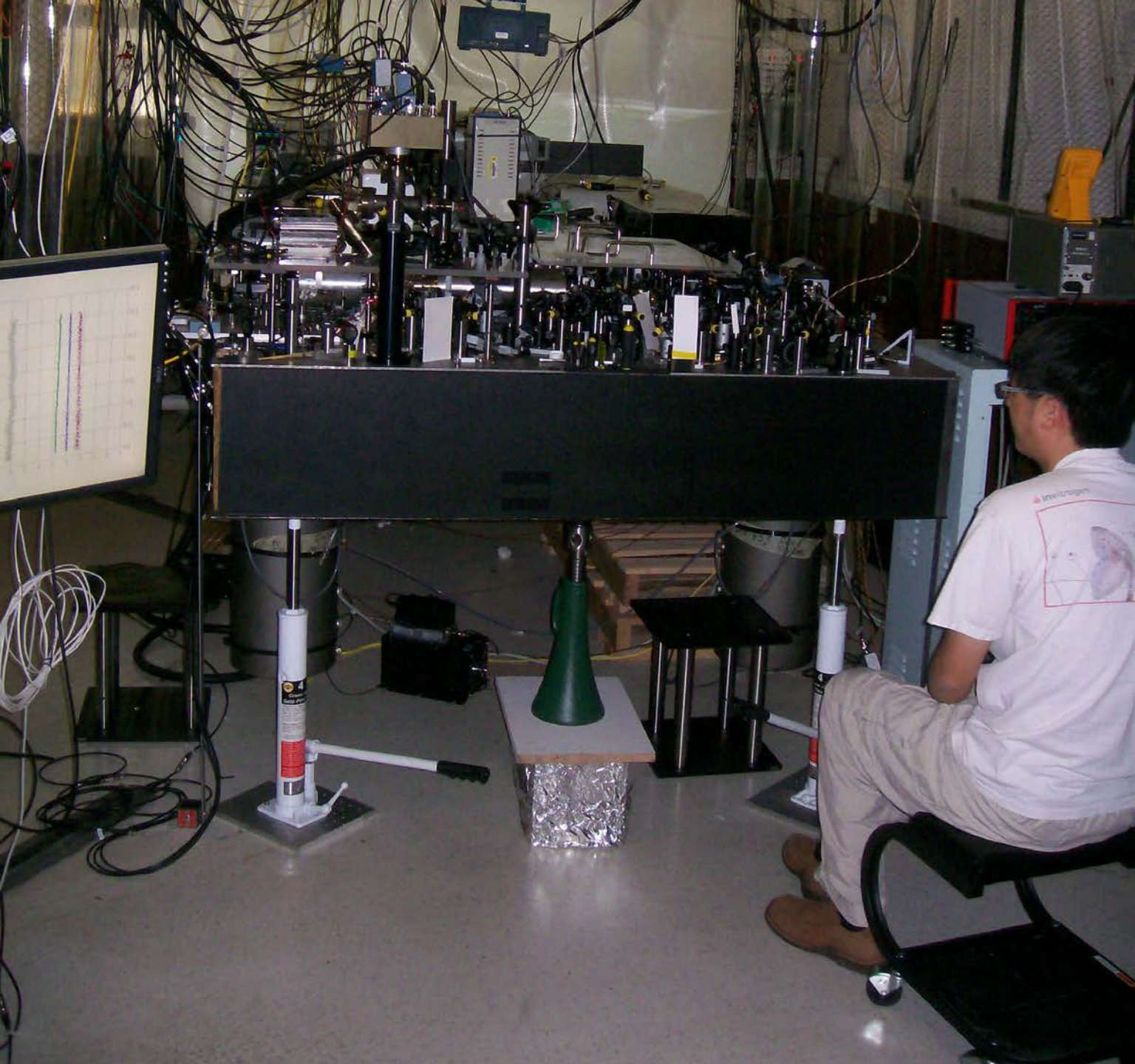
7×10^{-18}



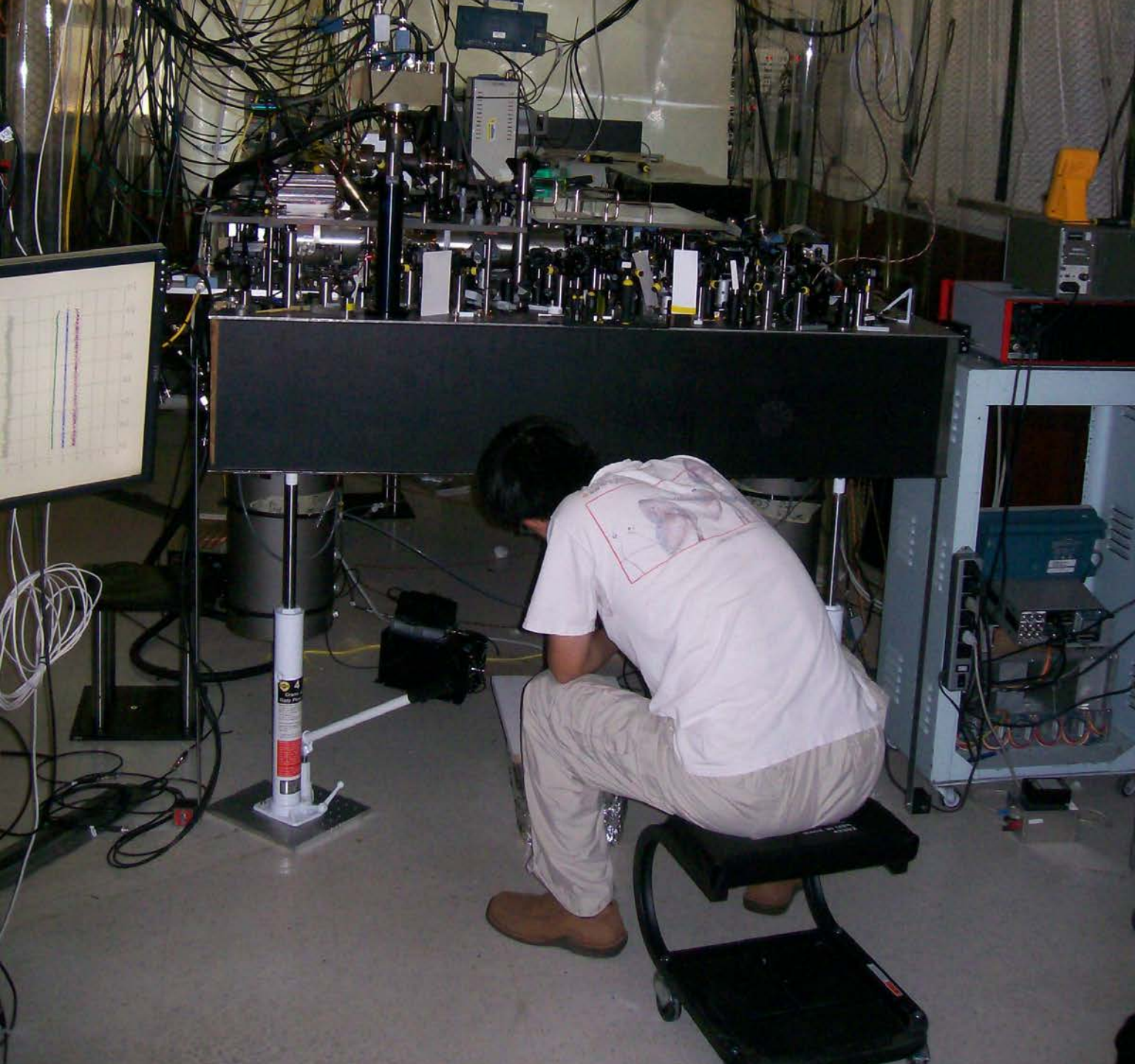
10×10^{-18}



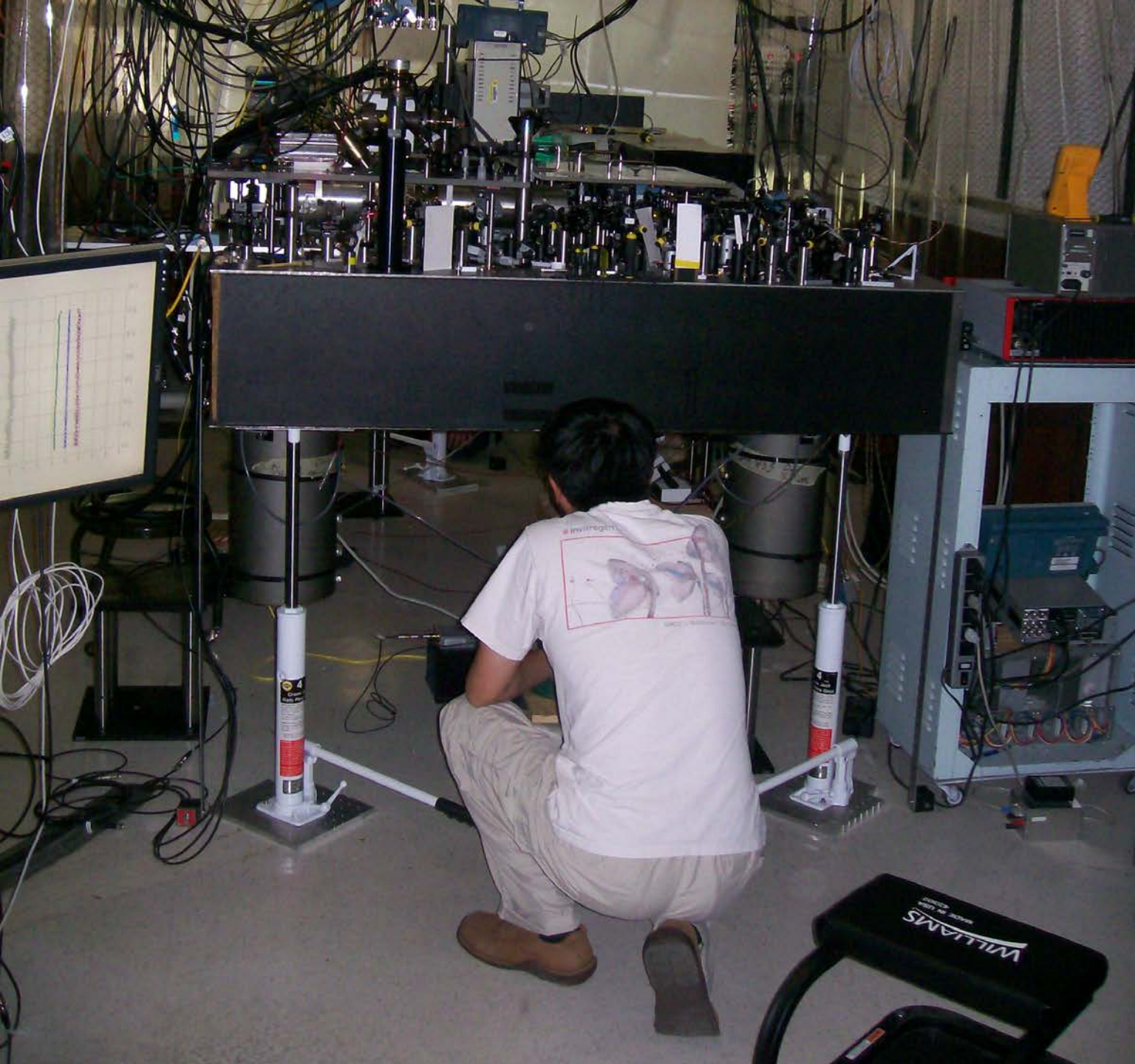
16×10^{-18}



20×10^{-18}



27×10^{-18}



33×10^{-18}

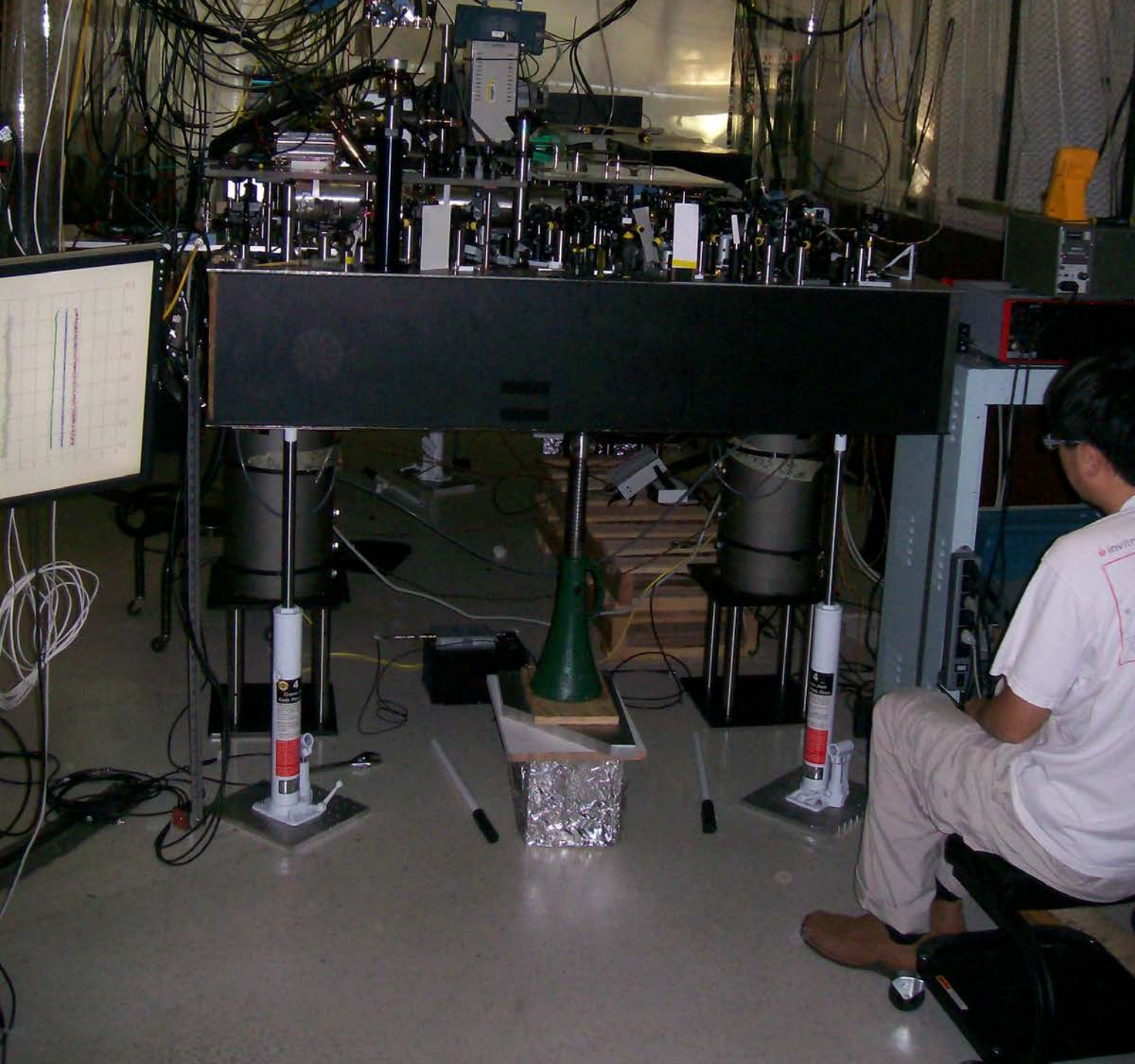


38×10^{-18}



38×10^{-18}

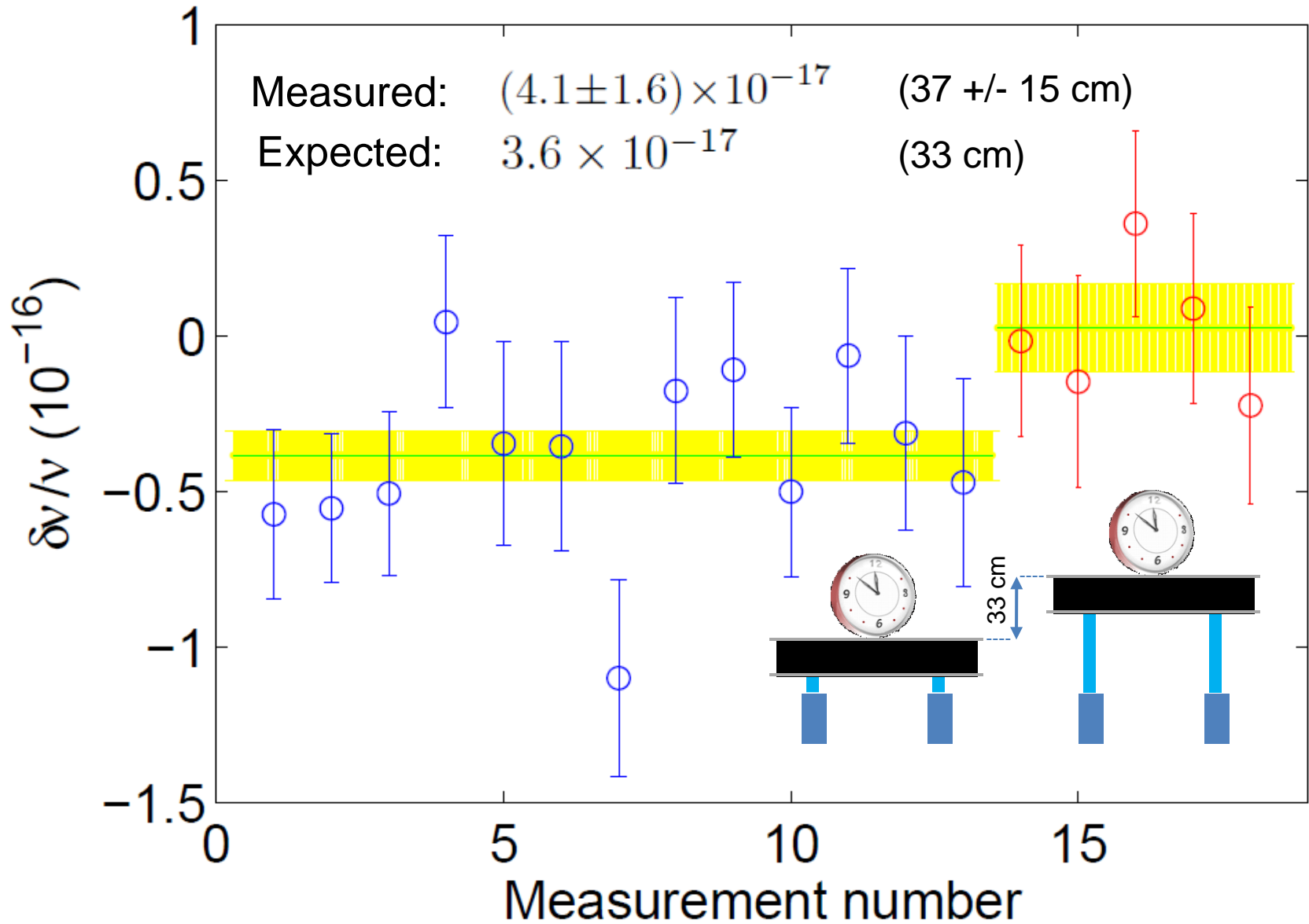
WILLIAMS
MADE IN USA



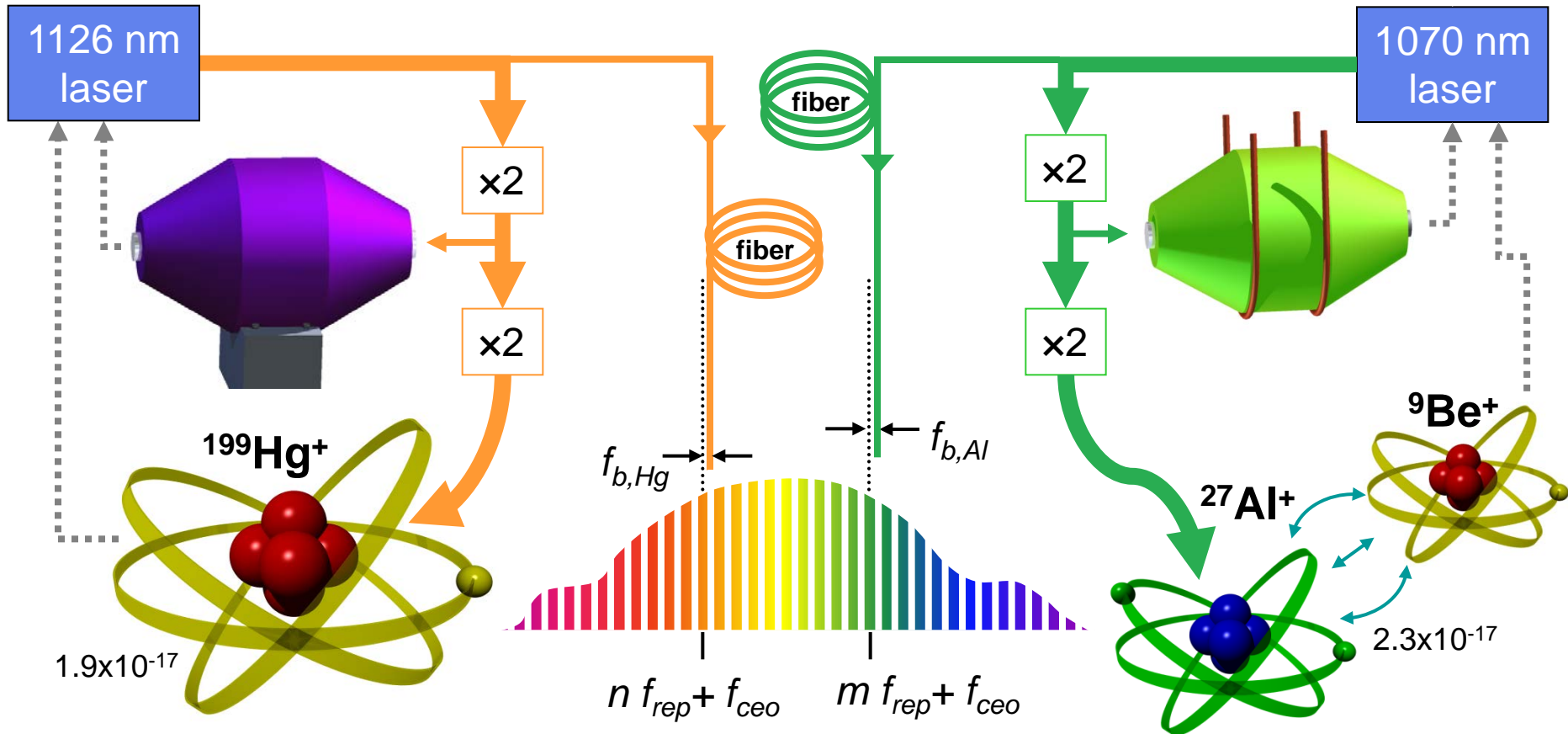
36×10^{-18}

Chou (front),
Hidden: Hume,
Rosenband

First sub-1 m height measurement with clocks



Al⁺/Hg⁺ Comparison



$$\frac{\nu_{\text{Al}^+}}{\nu_{\text{Hg}^+}} = 1.052\,871\,833\,148\,990\,438 \pm 5.5 \times 10^{-17}$$

Variation of fundamental constants



$$1.052\,871\,833\,148\,990\,438 \pm 5.5 \times 10^{-17} = r = \frac{\nu_{Al^+}}{\nu_{Hg^+}} = f\left(\alpha, \frac{m_p}{m_e}, g_p, g_n, \dots\right)$$

$$\alpha = \frac{e^2}{\hbar c} \approx \frac{1}{137.036}$$

Is α really constant?

Atomic structure calculations:

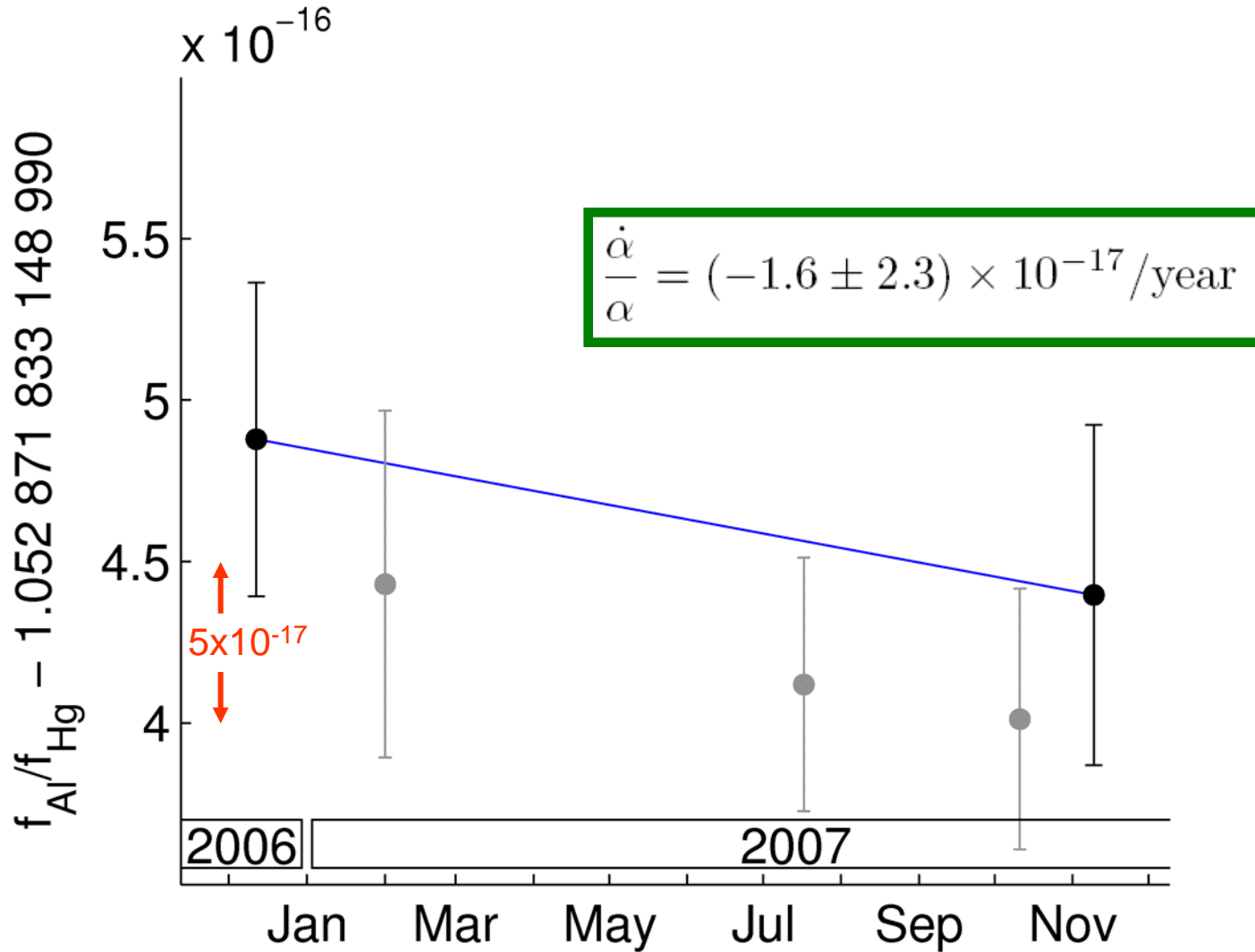
$$\frac{\Delta\alpha}{\alpha} \approx 0.34 \frac{\Delta r}{r}$$

V.A. Dzuba, V.V. Flambaum, J.K. Webb
Phys. Rev. A **49**, 230 (1999)

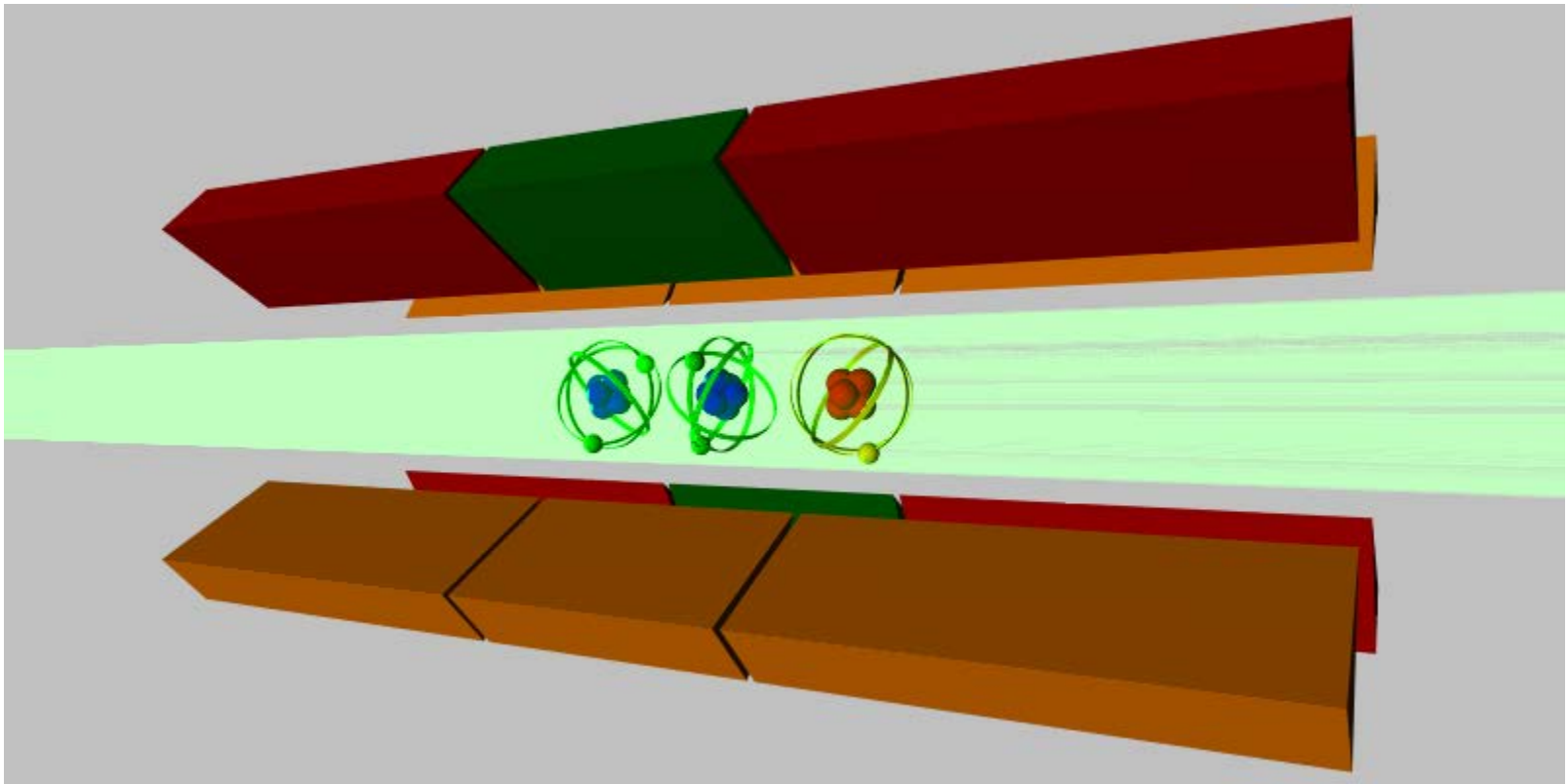
V.A. Dzuba, V.V. Flambaum
arXiv:0712.3621v1 (2008)

If α changes, then the frequency ratios r of different clocks will also change.

Al⁺/Hg⁺ Comparison (search for variation of α)

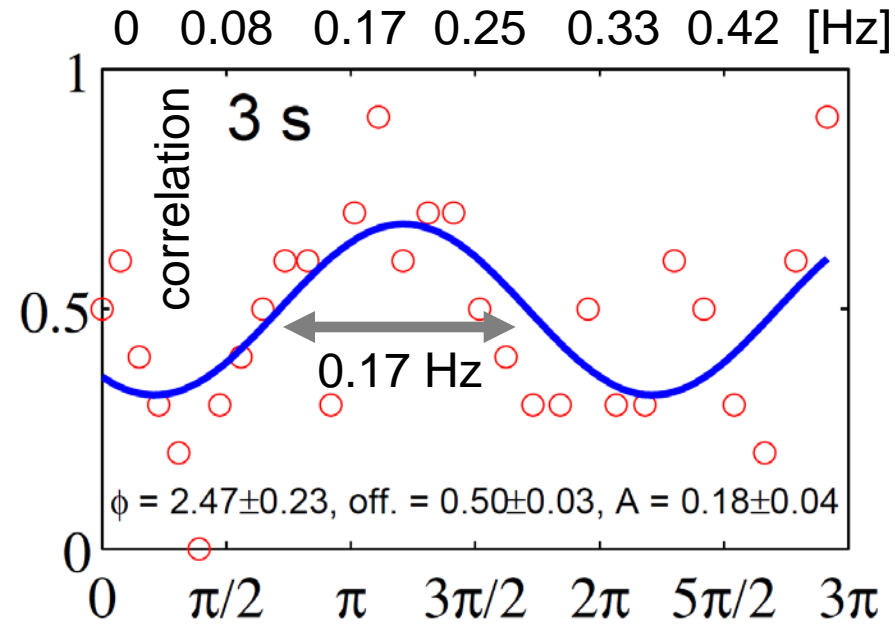
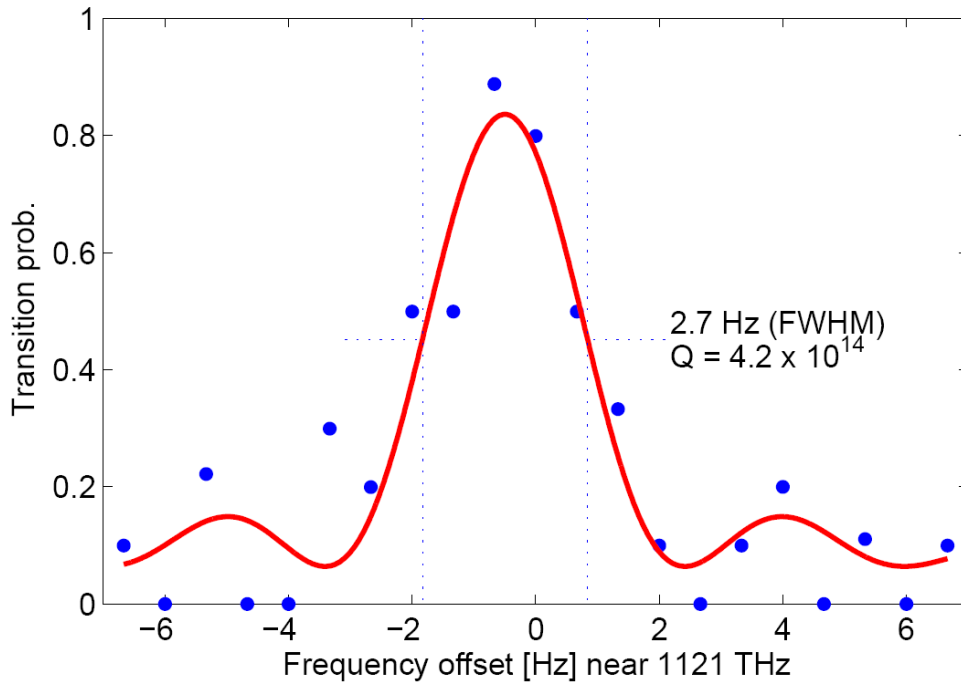


Quantum-correlation spectroscopy



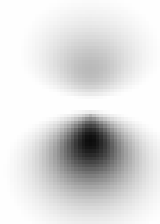
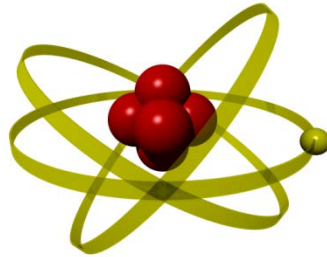
Factor 50-100 measurement speedup

- $$Q = \frac{1.121 \times 10^{15}}{0.17} = 6.7 \times 10^{15} \text{ (observed)}$$



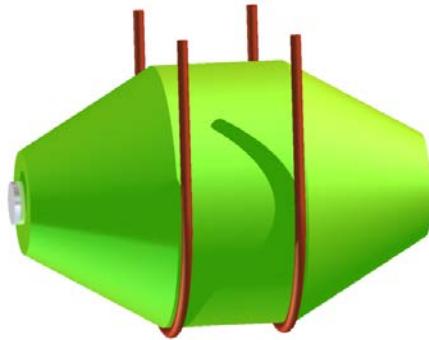
Quantum metrology in atomic clocks

Atoms



Very coherent.
Observation leads to
quantum projection
noise.

Classical oscillator



Less coherent.
Impervious to observation.

*How should we put these parts
together for best performance?*

arXiv: 1303.6357 [quant-ph]

Digitization of phase-difference
can yield exponential
performance gain compared to
simple averaging.

Summary

- $\text{Al}^+ / \text{Mg}^+$ clock has 8.6×10^{-18} accuracy
 $\approx \frac{1/100^{\text{th}} \text{ human hair diameter}}{\text{Earth-sun distance}}$
- Measured 37 +/- 15 cm height-difference via relativistic geodesy
- Constrained drift of fine-structure constant to $(-1.6 \pm 2.3) \times 10^{-17}/\text{year}$
- Quantum-correlation spectroscopy with Al^+ yields $Q > 6 \times 10^{15}$, Measurement speed-up of 50 – 100 x

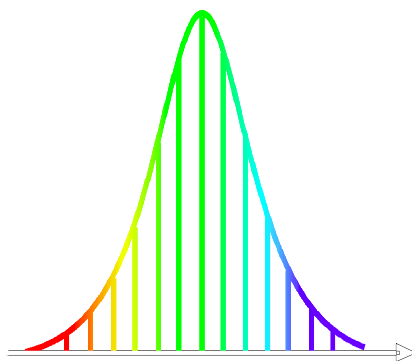
Acknowledgements

Al⁺

Jwo-Sy Chen
Sam Brewer
Chin-Wen Chou
David B. Hume
David J. Wineland

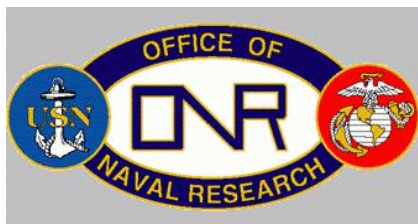
Hg⁺

David Leibrandt
David J. Wineland
James C. Bergquist



fs-comb (Ti:Sapphire)

Tara M. Fortier
Scott A. Diddams



NIST

